

1. For **each** of the following sets of sublevels, determine which has the lowest energy, and which has the highest energy.

(a) 2s, 3s, 4s

(b) 4f, 4p, 4d

(c) 4d, 4f, 3d

(d) 6s, 3d, 5f

Ex for (a) Lowest 2s, highest 4s

2. Explain what $6s^2$ means.

3. What is the maximum number of electrons that can be found in an s – sublevel? ____ p – sublevel? ____ d – sublevel? ____ f – sublevel? ____

4. Indicate which of these (there may be more than one) orbital designations is incorrect:

(a) 4s

(b) 3f

(c) 2d

(d) 3d

5. Draw energy level diagrams for

(a) sodium

(b) silicon

(c) beryllium

(d) cobalt

6. How many unpaired electrons are there in each of the atoms in question 5?

7. Give the electron configuration for:

(a) lithium

(b) oxygen

(c) selenium

(d) iodine

(e) silver

(f) arsenic

(g) indium

(h) potassium

8. Give the name and symbol for the atom whose electron configuration corresponds to each of the following electron configurations.

a) $1s^2 2s^2 2p^6 3s^2 3p^5$

b) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^3$

c) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^6$

d) $1s^2 2s^2 2p^6 3s^2 3p^2$

e) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$

9. Write the shorthand notation for the above electron configurations.

Electron Configuration Multiple Choice Questions

SA Adler



Electron Configuration Multiple Choice Questions:

This book delves into Electron Configuration Multiple Choice Questions. Electron Configuration Multiple Choice Questions is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Electron Configuration Multiple Choice Questions, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Electron Configuration Multiple Choice Questions
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 - Chapter 3: Electron Configuration Multiple Choice Questions in Everyday Life
 - Chapter 4: Electron Configuration Multiple Choice Questions in Specific Contexts
 - Chapter 5: Conclusion
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