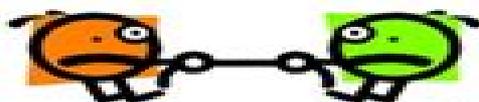


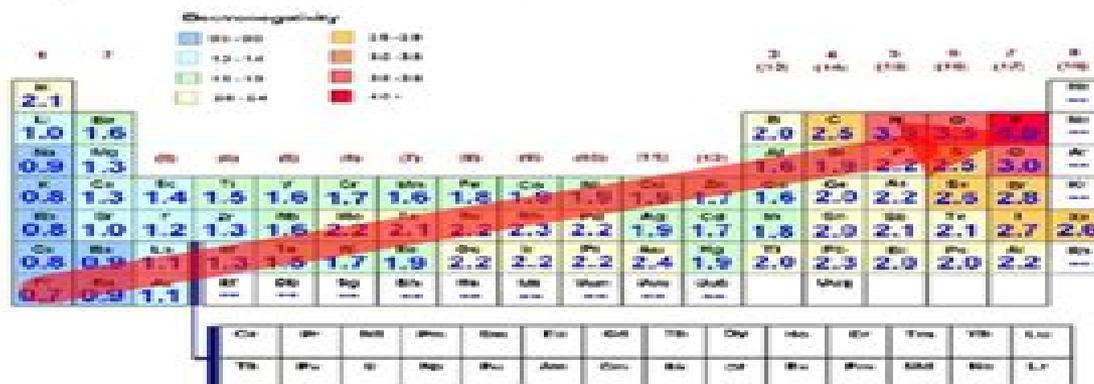
8.5 Electronegativity and Polarity

Electronegativity is the tendency of an atom to draw electrons closer to it **when it's in a chemical bond**.



If the rope in the picture represents electrons, the atom with the higher electronegativity will win the tug-of-war match for electrons.

Electronegativity increases as you move up and to the right on the periodic table towards F.



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J Spring



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