

Electrons in Atoms

Section 5.1 Light and Quantized Energy

In your textbook, read about the wave nature of light.

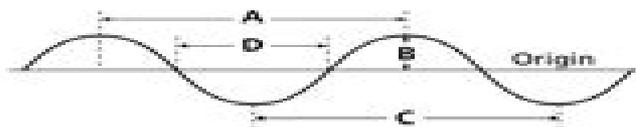
Use each of the terms below just once to complete the passage.

| | | | |
|-----------|--------|------------|-------|
| amplitude | energy | frequency | hertz |
| light | wave | wavelength | speed |

Electromagnetic radiation is a kind of **(1)** _____ that behaves like a(n) **(2)** _____ as it travels through space. **(3)** _____ is one type of electromagnetic radiation. Other examples include X rays, radio waves, and microwaves.

All waves can be characterized by their wavelength, amplitude, frequency, and **(4)** _____. The shortest distance between equivalent points on a continuous wave is called a(n) **(5)** _____. The height of a wave from the origin to a crest or from the origin to a trough is the **(6)** _____. **(7)** _____ is the number of waves that pass a given point in one second. The SI unit for frequency is the **(8)** _____, which is equivalent to one wave per second.

Use the figure to answer the following questions.



- Which letter(s) represent one wavelength?

- Which letter(s) represent the amplitude? _____
- If twice the length of A passes a stationary point every second, what is the frequency of the wave?

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Electrons In Atoms Chapter Test A Answer Key

J Spring



Electrons In Atoms Chapter Test A Answer Key:

Unveiling the Magic of Words: A Review of "**Electrons In Atoms Chapter Test A Answer Key**"

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