

Chapter 15 Alcohols Practice questions

Question number	Answer	Marks	Guidance	
1 (a)	A mixture of liquids is heated to boiling point for a prolonged time	1		
	Vapour is formed which escapes from the liquid mixture, is changed back into liquid and returned to the liquid mixture	1		
	Any ethanal and ethanol that initially evaporates can then be oxidised	1		
1 (b)	$\text{CH}_3\text{CH}_2\text{OH} + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{COOH} + 4\text{H}^+ + 4\text{e}^-$	1		
2 (a) (i)	Compounds with the same molecular formula	1	This is a definition so should be learnt.	
	But different structural formulae / different structures	1		
2 (a) (ii)	$\text{C}_2\text{H}_6\text{O}$ only	1		
2 (a) (iii)	CH_2 only	1		
2 (b)	Potassium dichromate(VI) / $\text{K}_2\text{Cr}_2\text{O}_7$ and acid / acidified / H_2SO_4 / H^+	1	<p>You can also have KMnO_4 / H_2SO_4, but not HCl.</p> <p>Or remain purple if KMnO_4 used.</p> <p>Or goes from purple to colourless if KMnO_4 in acid used or gives brown precipitate or goes green if KMnO_4 neutral or in alkali.</p>	
	Remains orange or no change or no reaction	1		
	Orange to green	1		
2 (c)	<u>Choice of reagents</u>			
	potassium dichromate (VI) / $\text{K}_2\text{Cr}_2\text{O}_7$ and acid / acidified / H_2SO_4 / H^+ or KMnO_4 / H_2SO_4	Fehling's / Benedict's reagent	Tollens' reagent or AgNO_3 / NH_3 or ammoniacal silver nitrate Not AgNO_3 alone.	1
	<u>with the aldehyde C</u>			
	goes orange to green goes purple to colourless / brown ppt. / green solution if KMnO_4 used red solid Potassium dichromate is more usual to do in school.	red solid	silver mirror	1
<u>with the ketone D</u>				
remains orange or no change or no reaction or purple for KMnO_4	remains blue or no change or no reaction	remains colourless or no change or no reaction	1	

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PT Brinkman



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