

**Directions:** Choose the letter that corresponds to the correct answer.

- Which of the following can be considered as Dobereiner's triad?
  - He: 4.0, Li: 6.9, Na: 23.0
  - N: 14.0, P: 40.0, At: 217.0
  - Ne: 20.2, Ar: 39.9, Kr: 83.8
  - Ca: 40.1, Sr: 87.6, Ba: 137.3
- Which of the following elements does not belong to the group?
  - N
  - S
  - Po
  - Te
- Which of the following is/are TRUE about boron and oxygen?
  - both are metalloids
  - boron is smaller than oxygen
  - boron is more electronegative than oxygen
  - higher energy is required to eject an electron from the valence shell of boron
  - all of the above
  - none of the above

For 4 and 5, consider Na, Mg, Al, and Si.

- The element that ejects one electron from its valence shell most readily is \_\_\_\_\_.
  - Na
  - Mg
  - Al
  - Si
- Which element has the most negative electron affinity?
  - Na
  - Mg

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web description publisher dc comics date published july 25 2018 part of scooby doo team up 40 of pages 20 writer sholly

fisch penciler dario brizuela inker dario brizuela colorer franco riesco letterer saida temofonte editor

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