

## Activity 1.2.1 Electron Theory – ANSWER KEY

### Introduction

The distribution of electrons in the orbital rings around an atom's nucleus determines an element's electrical properties. The stability of the electrons in the atom's outermost ring determine whether the element will act as a conductor, an insulator, or a semiconductor.

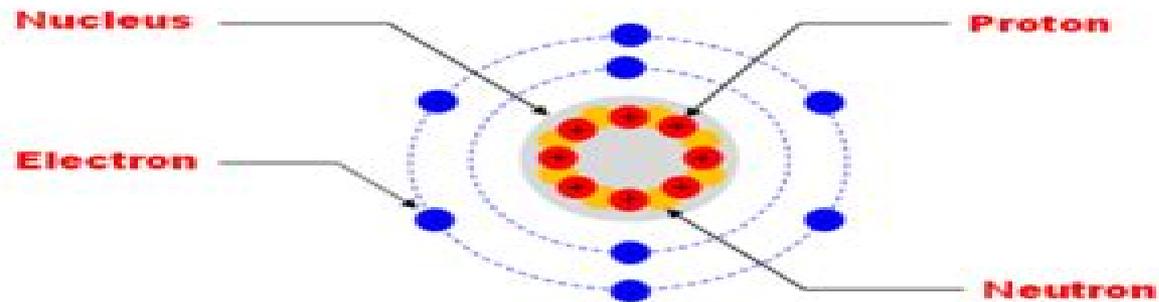
In this activity you will use the Internet to research the atomic structure of materials commonly used in electronic applications.

### Equipment

- Paper & pencil
- Computer (Internet access)

### Procedure

1. For the diagram shown below, label the parts of the atom. What element does this diagram represent? **Oxygen**



# Electron Theory Answer Guide

**Larry G. Sellers**



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