

EXP. 1 MEASUREMENT OF AN EQUILIBRIUM CONSTANT

POST-LAB REPORT WORKSHEETS

Concentration of standard $\text{Na}_2\text{S}_2\text{O}_3$ solution: 0.01105

Temperature of the equilibrium mixture: 21.1 °C

Part 1 – Titration of the Aqueous Layer

You must have 3 titrations.

Volume of Aqueous Aliquots:	1	2	3
Final Burette Reading:	<u>19.61</u>	<u>36.51</u>	<u>41.68</u>
Initial Burette Reading:	<u>10.54</u>	<u>20.53</u>	<u>30.21</u>
Volume of Aliquot:	<u>9.07</u>	<u>9.98</u>	<u>11.47</u>
Titration with Sodium Thiosulfate:			
Final Burette Reading:	<u>35.98</u>	<u>37.18</u>	<u>38.93</u>
Initial Burette Reading:	<u>3.34</u>	<u>4.88</u>	<u>4.81</u>
Volume of Titer:	<u>32.64</u>	<u>32.30</u>	<u>34.12</u>
Total Concentration of Reducible Iodine [$\text{I}_2 + \text{I}_3^-$] in Aqueous Phase:	<u>0.0199</u>	<u>0.0179</u>	<u>0.01644</u>

Equilibrium Constant Lab Report

LP Steffe



Equilibrium Constant Lab Report:

Equilibrium Constant Lab Report Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Equilibrium Constant Lab Report**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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