

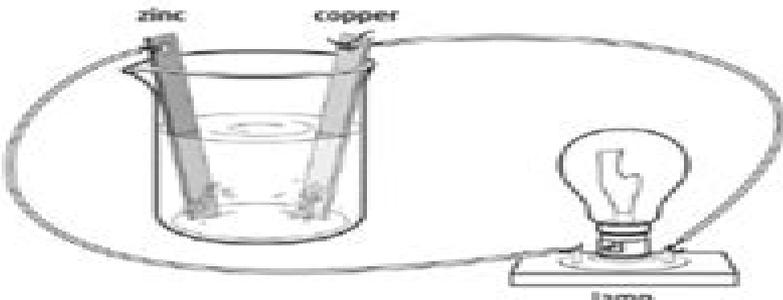
Chapter 11 Workbook

NAME: _____

Section 11.1 Review - Cells and Batteries

Multiple Choice

- Which two materials would make good electrodes for a voltaic (wet) cell?
 - copper and zinc
 - two strips of copper metal in distilled water
 - acid and copper
 - acid and zinc
 - two strips of zinc
- Which of the following is *not* an advantage of using a dry cell over a wet cell?
 - Dry cells are less easily damaged compared with wet cells.
 - Dry cells are easier to handle than voltaic cells.
 - Dry cells are used in portable radios.
 - Dry cells can be connected to make a battery, but wet cells cannot make a battery.
 - Dry cells can be transported more easily.
- What is a hybrid vehicle?
 - a vehicle that is fuelled by solar energy
 - a vehicle that is fuelled by solar energy and hydroelectric energy
 - a vehicle that is fuelled by two different types of fossil fuels
 - a vehicle that is fuelled by fossil fuels and electrical energy
 - a vehicle that is fuelled by propane only
- What types of cells can power a flashlight?
 - zinc-carbon cells
 - alkaline cells
 - rechargeable cells
 - only a voltaic wet cell
 - any of A, B, and C
- Which of the following is *true* for a voltaic cell?
 - The electrodes must be made of metal.
 - The electrolyte must be conducting.
 - The electrodes must be an insulator.
 - A and B are true.
 - B and C and true.
- Which cell has the least impact on the environment?
 - wet cell
 - dry cell
 - solar cell
 - fuel cell
 - alkaline cell
- What type of cell is a lead-acid cell?
 - primary dry cell
 - primary wet cell
 - secondary dry cell
 - secondary wet cell
 - none of the above
- Which of the following statements is *true* about the diagram below?



 - The zinc electrode forms positive ions, and electrons flow from zinc to copper through the lamp.
 - The zinc electrode forms negative ions, and electrons flow from zinc to copper through the lamp.
 - The zinc electrode forms positive ions, and electrons flow from copper to zinc through the lamp.
 - The zinc electrode forms positive ions, and protons flow from copper to zinc through the lamp.
 - The zinc electrode forms negative ions, and electrons flow from copper to zinc through the lamp.

Interactive Science Workbook Ch11

John S. Phillips, Cheryl Wistrom



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Table of Contents Interactive Science Workbook Ch11

1. Understanding the eBook Interactive Science Workbook Ch11
 - The Rise of Digital Reading Interactive Science Workbook Ch11
 - Advantages of eBooks Over Traditional Books
2. Identifying Interactive Science Workbook Ch11
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Interactive Science Workbook Ch11
 - User-Friendly Interface
4. Exploring eBook Recommendations from Interactive Science Workbook Ch11
 - Personalized Recommendations
 - Interactive Science Workbook Ch11 User Reviews and Ratings
 - Interactive Science Workbook Ch11 and Bestseller Lists
5. Accessing Interactive Science Workbook Ch11 Free and Paid eBooks
 - Interactive Science Workbook Ch11 Public Domain eBooks
 - Interactive Science Workbook Ch11 eBook Subscription Services
 - Interactive Science Workbook Ch11 Budget-Friendly Options

6. Navigating Interactive Science Workbook Ch11 eBook Formats
 - ePub, PDF, MOBI, and More
 - Interactive Science Workbook Ch11 Compatibility with Devices
 - Interactive Science Workbook Ch11 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Interactive Science Workbook Ch11
 - Highlighting and Note-Taking Interactive Science Workbook Ch11
 - Interactive Elements Interactive Science Workbook Ch11
8. Staying Engaged with Interactive Science Workbook Ch11
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Interactive Science Workbook Ch11
9. Balancing eBooks and Physical Books Interactive Science Workbook Ch11
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Interactive Science Workbook Ch11
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Interactive Science Workbook Ch11
 - Setting Reading Goals Interactive Science Workbook Ch11
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Interactive Science Workbook Ch11
 - Fact-Checking eBook Content of Interactive Science Workbook Ch11
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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