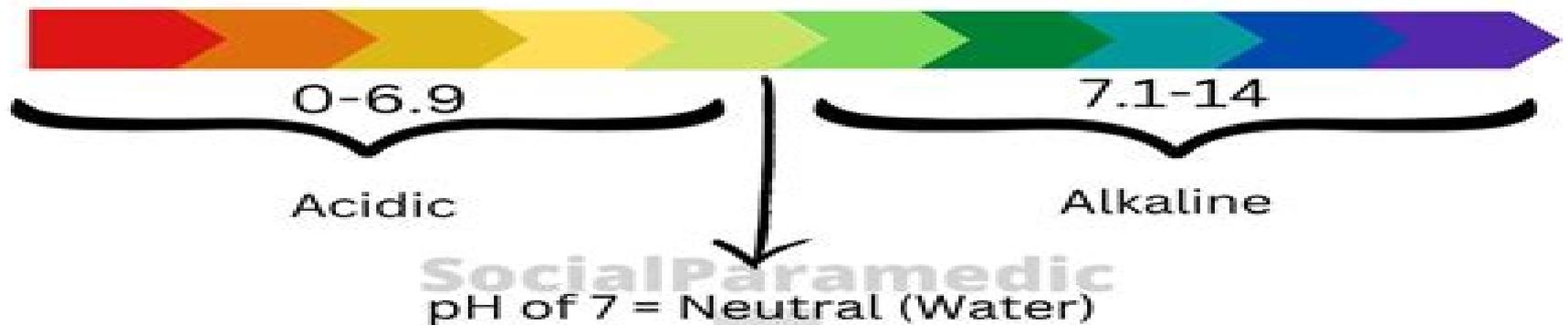


Acid-Base Balance

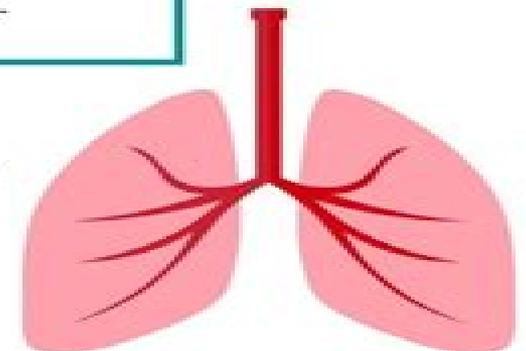


SocialParamedic
pH of 7 = Neutral (Water)

Normal Ranges

| | |
|-----------------------------------|-------------|
| pH of Blood | 7.35-7.45 |
| ETCO ₂ | 35-45mmHg |
| HCO ₃ (Bicarbonate) | 22-26 meq/L |

HCO₃: Normally, O₂ combines with glucose to create energy and CO₂ is an acid waste by product of this. Bicarbonate is a buffer produced by the body that is alkaline, and binds with CO₂ to neutralize it to allow the body to get rid of it.



Equilibrium And Acid Base Study Guide

Scott C. Dulebohn



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