
AQA A Level Chemistry

- Paper 3

what colour is $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$ - ANSWER pale green

Transition Metal definition - ANSWER D block elements that form at least one stable ion with a partially filled d orbital

Classic elements which are not transition metals despite being D-Block elements - ANSWER Scandium and Zinc

properties of transition metals which are NOT shared by other metals - ANSWER -
multiple oxidation states
-coloured ions in solution
-great catalysts
-form complexes with ligands

ligand - ANSWER a molecule which forms a dative bond with a transition metal

complex ion - ANSWER A transition metal ion bonded to one or more ligands by coordinate bonds

co-ordination number - ANSWER The number of co-ordinate bonds formed to a central metal ion

What is an example of incomplete ligand substitution? - ANSWER $[\text{Cu}(\text{NH}_3)_6]^{2+}$ to $[\text{Cu}(\text{NH}_3)_4(\text{H}_2\text{O})_2]^{2+}$

why does the coordination number not change when substituting water for ammonia ? - ANSWER because they are of a similar size

Why does incomplete ligand substitution usually occur? - ANSWER when substituting ligands of a different size

How To Answer Chemistry Paper 3

Rachel S Tattersall



How To Answer Chemistry Paper 3:

The Enigmatic Realm of **How To Answer Chemistry Paper 3**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **How To Answer Chemistry Paper 3** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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