

NAME: \_\_\_\_\_

**Reminders:**

1. In a neutral atom the number of protons equals the number of electrons.
2. An atom can NEVER gain or lose protons.
3. The number of protons equals the atomic number

# Ion Practice Set

1. What is an ion?
2. What does the number next to the ions signify?

*Complete the following table, using the periodic table in the back of your book.*

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	NUMBER OF ELECTRONS LOST OR GAINED
ex	Fluorine	F <sup>-</sup>	9	10	gained one
1			53	54	
2			16		gained two
3	potassium				lost one
4		Ca <sup>+2</sup>			
5			35	36	
6		Sr <sup>+2</sup>			
7		H <sup>+</sup>			
8			8		gained two
9			12		lost two
10	aluminum			10	
11			34	36	
12		H <sup>-</sup>			
13	lithium				lost one
14		Rb <sup>+</sup>			
15			17	18	

# Ion Practice Set Key

**L Towne**



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## Reviewing **Ion Practice Set Key**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Ion Practice Set Key**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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