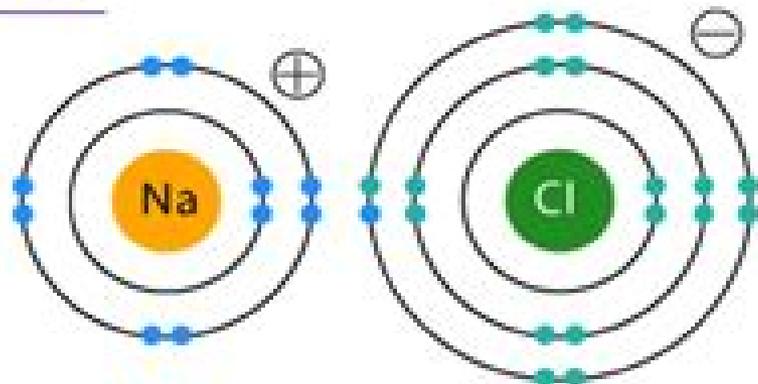


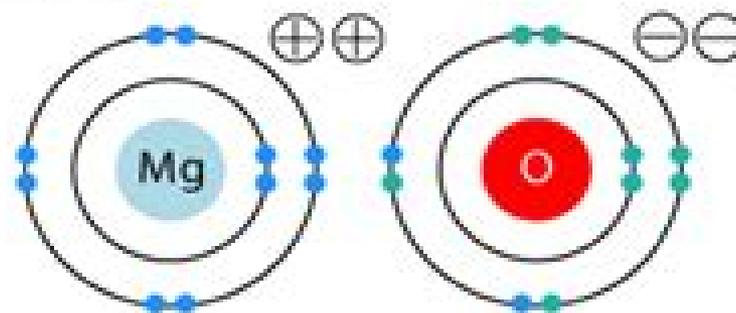
Naming Ionic Compounds

NaCl



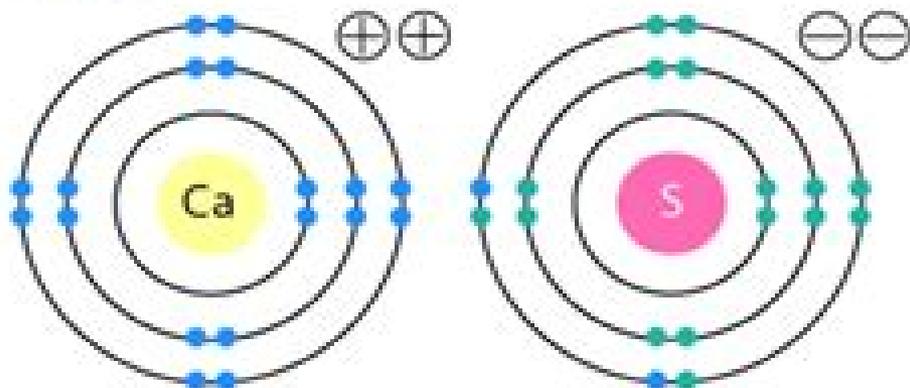
Sodium ion + Chloride ion = Sodium chloride

MgO



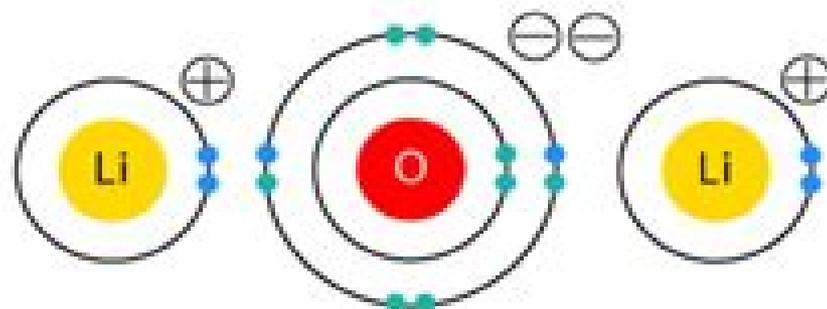
Magnesium ion + Oxide ion = Magnesium oxide

CaS



Calcium ion + Sulfide ion = Calcium sulfide

Li₂O



Lithium ions + Oxide ion = Lithium oxide

Ionic Compounds And Metals A Chemistry Handbook

Iris Stovall



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Aqa Chemistry for Gcse. Revision Guide Ben Clyde,2007-06-13 An exact match to AQA which includes personalised learning activities to enable students to review what they have learnt and advice from examiners on common pitfalls and how to avoid them

Longman Effective Guide to 'O' Level Chemistry Eng Wah Lim,2007-11-30 This book is specially written for students sitting for the Singapore Cambridge O Level Chemistry examination A comprehensive coverage of all the topics in the latest 2007 syllabus as well as mid year and final year examination papers enable students to study effectively and achieve success in their examinations

AQA GCSE 9-1 Chemistry Revision Guide: Ideal for the 2024 and 2025 exams (Collins GCSE Grade 9-1 Revision) Collins GCSE,2020-09-01 Exam Board AQA Level GCSE Grade 9 1 Subject Chemistry First Teaching September 2016 First Exams June 2018 Suitable for the 2020 autumn and 2021 summer exams

General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake,2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

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Study Guide General Chemistry, Second Edition, Becker/Wentworth Becker,Benjamin L. Carroll,1980

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Study Guide for Introductory Chemistry Iris Stovall,1990

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Fundamentals of Chemistry, Study Guide James E. Brady, John R. Holum, 1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs

Student's Guide to Brown and LeMay, Chemistry, the Central Science, 2nd Edition James C. Hill, 1981

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples

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Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

Study Guide for General Chemistry for Colleges Herman

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Hummel & Zumdahl, Steven S. Zumdahl,2004-08 A Laboratory Guide to the Study of Qualitative Analysis Edgar Henry
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