



IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each definition with the correct term. Write the letter of the correct term in the space provided.

Column I

- _____ 1. electrical energy
- _____ 2. ionization
- _____ 3. ionic bonding
- _____ 4. electron
- _____ 5. conductivity
- _____ 6. lattice ion
- _____ 7. atom
- _____ 8. compound
- _____ 9. crystalline
- _____ 10. noncrystalline
- _____ 11. crystalloid
- _____ 12. amorphous solid

Column II

- _____ 13. the attractive force between oppositely charged ions
- _____ 14. the attractive force between atoms, molecules, or ions
- _____ 15. the force between atoms, molecules, or ions
- _____ 16. the smallest particles of matter
- _____ 17. a substance that conducts heat well
- _____ 18. the smallest particles of matter
- _____ 19. an atom with a positive charge and a mass of 1 atomic mass unit
- _____ 20. the force between molecules of a substance
- _____ 21. the force between atoms of a substance
- _____ 22. the force between atoms of a substance
- _____ 23. the force between atoms of a substance
- _____ 24. the force between atoms of a substance
- _____ 25. the force between atoms of a substance
- _____ 26. the force between atoms of a substance
- _____ 27. the force between atoms of a substance
- _____ 28. the force between atoms of a substance
- _____ 29. the force between atoms of a substance
- _____ 30. the force between atoms of a substance

B. Multiple Choice

Write the letter of the correct answer in the space provided.

- _____ 1. Which of the following is not a characteristic of ionic bonding?
 - (a) high melting point
 - (b) brittleness
 - (c) electrical conductivity
 - (d) low boiling point

Ionic Metallic Bonding Answers Chapter Test

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