

Name: _____

Period: _____

Fall Semester Review – IPC

IPC Chemistry Review

Find the atomic mass of the following compounds.

Calcium oxide (CaO)

Water

Classify (ionic, covalent or polyatomic) and name the following compounds.

1. CO₂ _____
2. Na₂O _____
3. BeCrO₄ _____

Find the valence electrons and oxidation number of:

1. Helium _____
2. Oxygen _____

Draw the Lewis Dot Diagrams for:

1. Carbon
2. Sodium

Make Balanced Compounds from:

1. Li + O _____
2. Na + Ar _____
3. Be + NO₃⁻ _____
4. Ca + O _____

With dot diagrams draw the covalent bond of O₂.

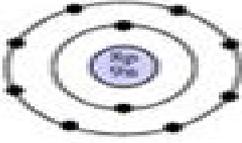
With dot diagrams the covalent bond of O₂.

Matching:

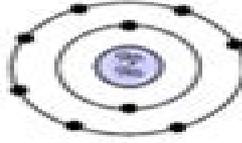
- A. Mixture
- B. Compound
- C. Element
- D. Matter
- E. Heterogeneous
- F. Homogeneous

- _____ A substance made up of two or more atoms that must be separated by chemical means.
- _____ Something made up of 2 or more substances that can be separated by physical means.
- _____ A substance that is pure and made up of only one type of atom.
- _____ Anything that has mass and takes up space.
- _____ A mixture that is the same throughout.
- _____ A mixture that is different throughout.

1. Valence electrons are the outermost electrons of an atom that are involved in chemical bonding? True/False.
2. Lithium has _____ valence electrons, will (lose or gain) _____ electrons and become (positive or negative). This is why lithium's oxidation number is _____.
3. A metal and non-metal will form a _____ compound, two non-metals form a _____ compound.
4. Protons are _____; electrons are _____; neutrons are _____.
5. Protons and neutrons are in the center of the atom, which is called the _____.



Atom A



Atom B

Which one is the ion?

Give the ion notation for the ion:

Give the number of protons for:

1. Carbon _____
2. Beryllium _____
3. Iron _____
4. Chlorine _____

Matching:

- A. Ion
- B. Element
- C. Isotope
- D. Oil

If you change the number of protons you change the _____.

If you change the number of neutrons, you change the _____.

If you change the number of electrons, you change the _____.

Every 5,000 miles you should change a car's _____.

Matching:

- A. Law of Conservation of Mass
- B. Meter
- C. Centimeter
- D. Liter
- E. Gram

_____ Mass is neither destroyed or created in chemical reactions.

_____ Unit of mass; about 1 dollar bill.

_____ 1/100th of meter; width of pinky finger.

_____ Unit of volume; just bigger a quart.

_____ Unit of length; 3.3.

Matching:

- A. Solid
- B. Liquid
- C. Gas
- D. Oil

_____ Molecules that are tightly packed and retain their shape and size.

_____ Molecules that bounce off of each other, can be compressed, and take the shape of their container.

_____ Molecules that can move (slide) over each other and have a definite size (volume), but not shape and can not be compressed.

IPC Chemistry Review

Daniel F McAuley



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