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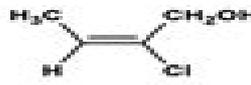
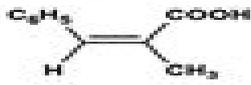
First Semester B.Tech Degree Regular and Supplementary Examination December 2020 (2019 scheme)

Course Code: CYT100**Course Name: ENGINEERING CHEMISTRY****(2019-Scheme)**

Max. Marks: 100

Duration: 3 Hours

PART A**Answer all questions, each carries 3 marks.**

- | | | Marks |
|----|--|-------|
| 1 | What will be the standard electrode potential of $\text{Ni}^{2+} / \text{Ni}$ electrode if the cell potential of the cell $\text{Ni} / \text{Ni}^{2+}(1\text{M}) // \text{Cu}^{2+}(0.1\text{M}) / \text{Cu}$ is 0.59 V at 25 °C?
$E^\circ_{\text{Cu}^{2+}/\text{Cu}} = 0.34 \text{ V}$ | (3) |
| 2 | Briefly explain the principle of electroless plating. | (3) |
| 3 | Give the mechanism of interaction of electromagnetic radiation with oscillating dipole. | (3) |
| 4 | State Beer-Lambert's law and write the differential form. | (3) |
| 5 | Write any three differences of TGA and DTA. | (3) |
| 6 | Explain sol-gel method for the synthesis of nano particles | (3) |
| 7 | Determine the configuration of the following alkenes as E or Z : | (3) |
| | <div style="display: flex; justify-content: space-around; align-items: center;"> <div style="text-align: center;">  <p>A</p> </div> <div style="text-align: center;">  <p>B</p> </div> </div> | |
| 8 | Mention any three advantages of OLEDs over LED and LCD. | (3) |
| 9 | Explain disinfection by chlorination. | (3) |
| 10 | Compare BOD and COD. | (3) |

PART B**Answer one full question from each module, each question carries 14 marks****Module-I**

- | | | |
|----|--|------|
| 11 | a) Explain the mechanism of electrochemical corrosion in different environmental conditions. | (10) |
| | b) How is the cell constant of a conductivity cell measured? | (4) |

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reactants b on the diagram provided in your answer booklet draw a dashed line to indicate a potential energy curve for the

reaction if a catalyst is added 46 given the potential energy diagram for a reaction

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web when the reactants have more potential energy than the products have and the extra energy is released to the

surroundings it is called activation energy the height of the hill on a potential energy diagram is a measure of the

endothermic the reaction above can be categorized as

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web this is like a one dimensional system whose mechanical energy e is a constant and whose potential energy with respect

to zero energy at zero displacement from the spring s unstretched length $x = 0$ is $u = \frac{1}{2} kx^2$ figure 8 5 2 8 5 2 a a glider

between springs on an air track is an example of a horizontal mass spring system

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web 1 if the pe of the reactants is 430kj and the pe of the products is 190kj what is the Δh 2 if the pe of the reactants is 430kj

and the activation energy is 220 how much total energy will be needed to be added for the reaction to proceed transcribed

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web key points potential energy is energy that has the potential to become another form of energy an object's potential energy depends on its physical properties and position in a system potential energy comes in many forms such as gravitational potential energy due to an object's mass and position in a gravitational field

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web the energy of the reactants is shown on the energy diagram above on the right side of the energy diagram draw a horizontal line segment to indicate the energy of the products draw a vertical line segment to indicate Δh for the reaction and label it with the correct value see attached image

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web aug 16 2019 1 answer sorted by 4 the activation energy is not the average kinetic energy that the colliding particles

need in order to successfully react instead it is the energy barrier between reactants and products often the energy required to cross the barrier comes from the kinetic energy of collisions

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