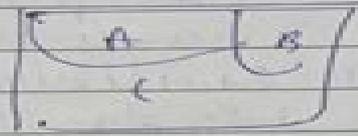


IUPAC Sign Convention

Work done on the system = +ve work
Work done by the system = -ve work

Heat given to the system = +ve heat ($+q$)
Heat given by the system = -ve heat ($-q$)

Zeroth Law



$$\begin{aligned} &T_A = T_B \\ &T_B = T_C \end{aligned} \Rightarrow T_A = T_C$$

First Law of Thermodynamics

Energy of the universe is always constant. The
law of conservation of energy is universal.

Energy cannot be created or destroyed, it only
changes form.

Law of conservation of energy is an application of the
law of conservation of mass.



Heat is a form of energy.

Fiitjee Solved File Solution Of Iit Jeemains Paper 2014

Yan Bai



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