



Titration

Answer Key

Vocabulary: acid, analyte, base, dissociate, equivalence point, indicator, litmus paper, molarity, neutralize, pH, strong acid, strong base, titrant, titration, titration curve, weak acid, weak base

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

[Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

There are several definitions of acids and bases. According to the Brønsted-Lowry definition, an **acid** is a substance that is capable of donating a proton to another substance. A **base** is a substance that accepts protons. When an acid and a base are combined, the acid is **neutralized** as the base accepts the protons produced by the acid.



One way to determine if a solution is acidic or basic is to use **litmus paper**, as shown above. There are two types of litmus papers: red and blue.

How does litmus paper indicate an acid? *Both strips turn red.*

How does litmus paper indicate a neutral substance? *The red strip stays red and the blue strip stays blue.*

How does litmus paper indicate a base? *Both strips turn blue.*

Gizmo Warm-up

Litmus is an example of an **indicator**, a substance that changes color depending on its **pH** (pH is a measure of the concentration of protons, or H^+ ions). In the *Titration Gizmo*, you will use indicators to show how acids are neutralized by bases, and vice versa.

To begin, check that **1.00 M NaOH** is selected for the **Burette**, **Mystery HBr** is selected for the **Flask**, and **Bromthymol blue** is selected for the **Indicator**.



1. Look at the flask. What is the color of the bromthymol blue indicator? *Yellow*

2. What does this tell you about the pH of the solution in the flask? *pH is below 6.0*

Solutions with a pH below 7.0 are acidic, while those with a pH above 7.0 are basic.

3. Move the slider on the burette to the top to add about 25 mL of NaOH to the flask. What happens, and what does this tell you about the pH of the flask?

The solution turns blue, indicating the pH is now above 7.6.



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George Fuller



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Table of Contents Gizmo Answers Key For

1. Understanding the eBook Gizmo Answers Key For
 - The Rise of Digital Reading Gizmo Answers Key For
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answers Key For
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Answers Key For
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Answers Key For
 - Personalized Recommendations
 - Gizmo Answers Key For User Reviews and Ratings
 - Gizmo Answers Key For and Bestseller Lists
5. Accessing Gizmo Answers Key For Free and Paid eBooks
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 - Gizmo Answers Key For Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Answers Key For
 - Highlighting and Note-Taking Gizmo Answers Key For
 - Interactive Elements Gizmo Answers Key For
- 8. Staying Engaged with Gizmo Answers Key For
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Answers Key For
- 9. Balancing eBooks and Physical Books Gizmo Answers Key For
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Answers Key For
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Answers Key For
 - Setting Reading Goals Gizmo Answers Key For
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Answers Key For
 - Fact-Checking eBook Content of Gizmo Answers Key For
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
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