

Name : \_\_\_\_\_ Date : \_\_\_\_\_

# Lab Periodic Trends Worksheet

The atomic and the ionic radius of the first 18 elements of the periodic table are given below.

| Element           | H     | He   | Li   | Be   | B    | C    | N    | O    | F    |
|-------------------|-------|------|------|------|------|------|------|------|------|
| Atomic Number     | 1     | 2    | 3    | 4    | 5    | 6    | 7    | 8    | 9    |
| Atomic Radius (Å) | 0.79  | 0.49 | 2.05 | 1.4  | 1.17 | 0.91 | 0.75 | 0.65 | 0.57 |
| Ionic Radius (Å)  | 0.012 | -    | 0.76 | 0.35 | 0.23 | 0.16 | 0.13 | 1.4  | 1.33 |

| Ne   | Na   | Mg   | Al    | Si   | P    | S    | Cl   | Ar   |
|------|------|------|-------|------|------|------|------|------|
| 10   | 11   | 12   | 13    | 14   | 15   | 16   | 17   | 18   |
| 0.38 | 2.23 | 1.72 | 1.82  | 1.46 | 1.23 | 1.09 | 0.97 | 0.71 |
| -    | 1.02 | 0.72 | 0.535 | 0.4  | 0.38 | 0.37 | 1.81 | -    |

Using the data above, construct the line graph of atomic radius vs. atomic number and ionic radius vs. the atomic number.

**Atomic Radius vs. Atomic Number**



**Ionic Radius vs. Atomic Number**



# Lab Periodic Trends Computer Activity Answer Key

**ML Morrison**



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## **Table of Contents Lab Periodic Trends Computer Activity Answer Key**

1. Understanding the eBook Lab Periodic Trends Computer Activity Answer Key
  - The Rise of Digital Reading Lab Periodic Trends Computer Activity Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Lab Periodic Trends Computer Activity Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Lab Periodic Trends Computer Activity Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Lab Periodic Trends Computer Activity Answer Key
  - Personalized Recommendations

- Lab Periodic Trends Computer Activity Answer Key User Reviews and Ratings
- Lab Periodic Trends Computer Activity Answer Key and Bestseller Lists
- 5. Accessing Lab Periodic Trends Computer Activity Answer Key Free and Paid eBooks
  - Lab Periodic Trends Computer Activity Answer Key Public Domain eBooks
  - Lab Periodic Trends Computer Activity Answer Key eBook Subscription Services
  - Lab Periodic Trends Computer Activity Answer Key Budget-Friendly Options
- 6. Navigating Lab Periodic Trends Computer Activity Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Lab Periodic Trends Computer Activity Answer Key Compatibility with Devices
  - Lab Periodic Trends Computer Activity Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Lab Periodic Trends Computer Activity Answer Key
  - Highlighting and Note-Taking Lab Periodic Trends Computer Activity Answer Key
  - Interactive Elements Lab Periodic Trends Computer Activity Answer Key
- 8. Staying Engaged with Lab Periodic Trends Computer Activity Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Lab Periodic Trends Computer Activity Answer Key
- 9. Balancing eBooks and Physical Books Lab Periodic Trends Computer Activity Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Lab Periodic Trends Computer Activity Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Lab Periodic Trends Computer Activity Answer Key
  - Setting Reading Goals Lab Periodic Trends Computer Activity Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Lab Periodic Trends Computer Activity Answer Key
  - Fact-Checking eBook Content of Lab Periodic Trends Computer Activity Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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