



# 6

## Chapter 6 Test, Form 1

NAME \_\_\_\_\_

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Write the letter for the correct answer in the blank at the right of each question.

1. There are 15 plums and 9 apples in a fruit bowl. Find the ratio of apples to plums. 1. \_\_\_\_\_  
 A. 3:5                      B. 3:8                      C. 5:3                      D. 8:3

2. The scale drawing of a porch is 8 inches wide by 12 inches long. If the actual porch is 12 feet wide, find the length of the porch. 2. \_\_\_\_\_  
 A. 8 ft                      B. 10 ft                      C. 16 ft                      D. 18 ft

3. Solve  $\frac{5}{2} = \frac{3}{x}$ . 3. \_\_\_\_\_  
 A. 4.6                      B. 4.8                      C. 5                      D. 7

4. A quality control technician checked a sample of 30 bulbs. Two of the bulbs were defective. If the sample was representative, find the number of bulbs expected to be defective in a case of 450. 4. \_\_\_\_\_  
 A. 24                      B. 30                      C. 36                      D. 45

5. Find the triangle similar to  $\triangle ABC$  at the right. 5. \_\_\_\_\_

A.



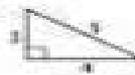
B.



C.



D.



6. Find  $x$  if  $\triangle ABC \sim \triangle JKL$ . 6. \_\_\_\_\_  
 A. 10                      B. 14  
 C. 25                      D. 29



7. Quadrilateral  $ABCD \sim$  quadrilateral  $PQRS$ . If  $AB = 10$ ,  $BC = 6$ ,  $PS = 12$ , and  $QR = 4$ , find the scale factor of  $ABCD$  to  $PQRS$ . 7. \_\_\_\_\_

A.  $\frac{1}{2}$

B.  $\frac{3}{2}$

C.  $\frac{5}{3}$

D.  $\frac{5}{6}$

8. If quadrilateral  $ABCD \sim$  quadrilateral  $EFGH$ , find  $x$ . 8. \_\_\_\_\_

A. 15

B. 20

C. 25

D. 30



9. Which theorem or postulate can be used to prove that these two triangles are similar? 9. \_\_\_\_\_

A. AA

B. SAS

C. SSA

D. SSS



10. Find  $MN$ . 10. \_\_\_\_\_

A.  $5\frac{1}{2}$

B.  $6\frac{3}{4}$

C. 7

D. 12



Assessments

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a hydroxide ion  $\text{OH}^-$  which is a polyatomic anion

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web oct 13 2023 this means that you will need to balance the carbon atoms first 5 use a coefficient to balance the single carbon atom add a coefficient to the single carbon atom on the right of the equation to balance it with the 3 carbon atoms on the left of the equation  $\text{C}_3\text{H}_8\text{O}_2 \rightarrow 2\text{H}_2\text{O} + 3\text{CO}_2$