

Gas Stoichiometry Worksheet

Name:

Solve all the following gas law problems. Show all work, answers are given at the end of the problem.

Molar Volume

1. Calculate the number of moles contained in 550.mL of carbon dioxide at STP. (0.0246mol)

2. Calculate the mass of 1.50 L of CH_4 at STP. (1.07g)

3. Calculate the volume in liters of 50.0 grams of nitrogen dioxide at STP. (24.3L)

Gas Stoichiometry

4. Given the following reaction: $\text{NH}_4\text{NO}_2 (\text{s}) \rightarrow \text{N}_2 (\text{g}) + 2\text{H}_2\text{O} (\text{g})$ How many liters of nitrogen gas is produced if 50.0L of water is produced at STP? (25.0L)

5. Given the following reaction: $\text{CuO} (\text{s}) + \text{H}_2 (\text{g}) \rightarrow \text{Cu} (\text{s}) + \text{H}_2\text{O} (\text{g})$ If 250. L of hydrogen gas are used to reduce copper (II) oxide at STP, what mass of copper is to be expected? (709g)

6. How many grams of solid lithium must be added to liquid water in order to obtain 15.0 L of hydrogen gas at STP? (9.29g) $2\text{Li} + 2\text{H}_2\text{O} \rightarrow 2\text{LiOH} + \text{H}_2$

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LM Reder

A decorative graphic consisting of a horizontal light blue bar with a rounded right end, and a red circular glow behind it.

Gas Stoichiometry Practice Test:

Whispering the Techniques of Language: An Psychological Quest through **Gas Stoichiometry Practice Test**

In a digitally-driven world wherever screens reign great and immediate transmission drowns out the subtleties of language, the profound strategies and mental nuances hidden within words usually move unheard. However, situated within the pages of **Gas Stoichiometry Practice Test** a interesting fictional value sporting with natural feelings, lies a fantastic journey waiting to be undertaken. Penned by a talented wordsmith, this enchanting opus attracts viewers on an introspective journey, delicately unraveling the veiled truths and profound influence resonating within the very cloth of each and every word. Within the psychological depths of this moving evaluation, we will embark upon a heartfelt exploration of the book is key subjects, dissect its captivating publishing type, and yield to the strong resonance it evokes serious within the recesses of readers hearts.

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