

General Instructions :

Read the following instructions very carefully and strictly follow them :

- (i) This question paper comprises **15** questions. **All** questions are compulsory.
- (ii) This question paper is divided into **three** sections – **A, B** and **C**.
- (iii) **Section A** – Questions No. **1** to **7** are short answer type questions. Each question carries **2** marks.
- (iv) **Section B** – Questions No. **8** to **13** are also short answer type questions. Each question carries **3** marks.
- (v) **Section C** – Questions No. **14** and **15** are case-based questions. Each question carries **4** marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION A

1. (a) Write the molecular formula of the following carbon compounds :
 - (i) Methane
 - (ii) Propane(b) Carbon compounds have low melting and boiling points. Why ? 2
2. The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
 - (a) Write the group numbers of these elements in the Modern Periodic Table.
 - (b) Write the molecular formula of the compound formed when X and Y combine with each other.
 - (c) Which of the two is electropositive ? 2
3. (a) Which of the following flowers will have higher possibility of self-pollination ?

Mustard, Papaya, Watermelon, Hibiscus

(b) List the two reproductive parts of a bisexual flower. 2
4. Which one of the two multicellular organisms — Spirogyra and Planaria reproduces by regeneration and why ? Give an example of any other organism which can also reproduce by the same process. 2



Grade 1march 2caps Question Paper

Rosina Ehmann



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Creamery and Milk Plant Monthly, 1939 Mekeel's Weekly Stamp News I. A. Mekeel, Charles Esterly Severn, Stephen B. Hopkins, 1913 Webster's New International Dictionary of the English Language, Based on the International Dictionary 1890 and 1900 William Torrey Harris, Frederic Sturges Allen, 1911 The Milk Dealer, 1951 **Random House Webster's College Dictionary** Random House (Firm), 1999

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