

Ionic Compounds and Formula Worksheet

Write the formula for each of the following compounds.

1. Magnesium oxide	MgO	2. Sodium fluoride	NaF
3. Aluminum nitride	Al₂N₃	4. Potassium sulfide	K₂S
5. Lithium iodide	LiI	6. Calcium bromide	CaBr₂
7. Beryllium oxide	BeO	8. Nickel chloride	NiCl
9. Magnesium nitride	Mg₃N₂	10. Aluminum sulfide	Al₂S₃
11. Calcium phosphide	Ca₃P₂	12. Zinc fluoride	ZnF
13. Sodium oxide	Na₂O	14. Lithium chloride	LiCl
15. Zinc oxide	ZnO	16. Silver chloride	AgCl
17. Barium bromide	BaBr₂	18. Potassium phosphide	K₃P
19. Silver oxide	Ag₂O		

Write the names for the following compounds.

1. Li ₂ O	Lithium oxide	2. CaF ₂	Calcium fluoride
3. AlCl ₃	Aluminum chloride	4. K ₃ P	Potassium phosphide
5. MgS	Magnesium sulfide	6. ZnF ₂	Zinc fluoride
7. CaO	Calcium oxide	8. AgCl	Silver chloride
9. KBr	Potassium bromide	10. Al ₂ S ₃	Aluminum sulfide
11. BeF	Beryllium fluoride	12. BaBr ₂	Barium bromide
13. Na ₃ N	Sodium nitride	14. CaS	Calcium sulfide
15. Al ₂ O ₃	Aluminum oxide	16. KI	Potassium iodide
17. Na ₂ S	Sodium sulfide	18. ZnO	Zinc oxide
19. Mg ₃ P ₂	Magnesium phosphide		

Formulae Of Ionic Compounds Answer Key

SA Dillow



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Formulae Of Ionic Compounds Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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