

General Chemistry – Unit 1 Worksheet 6

Dimensional Analysis

Use the factor-label method to make the following conversions. Remember to use the appropriate number of sf's in your answer.

Part 1

1. $74 \text{ cm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ meters
2. $8.32 \times 10^{-2} \text{ kg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ grams
3. $55.5 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}} \text{ cm}^3$
4. $0.00527 \text{ cal} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ kilocalories
5. $9.52 \times 10^{-4} \text{ m} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ micrometers
6. $41.0 \text{ mL} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ liters
7. $6.0 \times 10^{-1} \text{ g} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ mg
8. $8.34 \times 10^{-9} \text{ cg} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ g
9. $5.0 \times 10^3 \text{ mm} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ m
10. $1 \text{ day} \times \underline{\hspace{1cm}} \times \underline{\hspace{1cm}} \times \underline{\hspace{1cm}} = \underline{\hspace{2cm}}$ seconds
11. $5 \times 10^4 \text{ mm} \times \underline{\hspace{1cm}} \times \underline{\hspace{1cm}} = \underline{\hspace{2cm}}$ km
12. $9.1 \times 10^{-13} \text{ kg} \times \underline{\hspace{1cm}} \times \underline{\hspace{1cm}} = \underline{\hspace{2cm}}$ ng
13. $1 \text{ year} \times \underline{\hspace{2cm}} \times \underline{\hspace{2cm}} = \underline{\hspace{2cm}}$ hours (approximately)

General Chemistry Answer Key

Camilla Rothe



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Table of Contents General Chemistry Answer Key

1. Understanding the eBook General Chemistry Answer Key
 - The Rise of Digital Reading General Chemistry Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying General Chemistry Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an General Chemistry Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from General Chemistry Answer Key
 - Personalized Recommendations
 - General Chemistry Answer Key User Reviews and Ratings
 - General Chemistry Answer Key and Bestseller Lists
5. Accessing General Chemistry Answer Key Free and Paid eBooks
 - General Chemistry Answer Key Public Domain eBooks
 - General Chemistry Answer Key eBook Subscription Services

- General Chemistry Answer Key Budget-Friendly Options
- 6. Navigating General Chemistry Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - General Chemistry Answer Key Compatibility with Devices
 - General Chemistry Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of General Chemistry Answer Key
 - Highlighting and Note-Taking General Chemistry Answer Key
 - Interactive Elements General Chemistry Answer Key
- 8. Staying Engaged with General Chemistry Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers General Chemistry Answer Key
- 9. Balancing eBooks and Physical Books General Chemistry Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection General Chemistry Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine General Chemistry Answer Key
 - Setting Reading Goals General Chemistry Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of General Chemistry Answer Key
 - Fact-Checking eBook Content of General Chemistry Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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