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Measurements in Chemistry

Chapter 2

Problem-Set Solutions

- 2.1 It is easier to use because it is a decimal unit system.
- 2.2 Common measurements include mass, volume, length, time, temperature, pressure, and concentration.
- 2.3 a. The metric prefix *giga*, abbreviated as G, has a value of 10^9 . b. The metric prefix *nano*, abbreviated as n, has a value of 10^{-9} . c. The metric prefix *mega*, abbreviated as M, has a value of 10^6 . d. The metric prefix *micro*, abbreviated as μ , has a value of 10^{-6} .
- 2.4 a. kilo-, 10^3 b. M, 10^6 c. pico-, p d. d, 10^{-1}
- 2.5 a. A kilogram, abbreviated as kg, measures mass.
b. A megameter, abbreviated as Mm, measures length.
c. A nanogram, abbreviated as ng, measures mass.
d. A milliliter, abbreviated as mL, measures volume.
- 2.6 a. centimeter, length b. volume, dL c. picometer, length d. mass, kg
- 2.7 a. A milliliter is 10^3 times smaller than a liter.
b. A kiloliter is 10^3 times larger than a microliter. c. A nanoliter is 10^9 times smaller than a deciliter. d. A centiliter is 10^2 times smaller than a megaliter.
- 2.8 a. larger, 10^6 b. smaller, 10^3 c. smaller, 10^{10} d. smaller, 10^{15}
- 2.9 The meaning of a metric system prefix is independent of the base unit it modifies. The lists, arranged from smallest to largest are:
a. nanogram, milligram, centigram b. kilometer, megameter, gigameter
c. picoliter, microliter, deciliter d. microgram, milligram, kilogram

General Chemistry 6th Edition Solution Manual

Darrell D. Ebbing



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