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# ACS Gen Chem Ch. 1 Conceptual

## verified question with 100% answers

1. A sodium ion differs from a sodium ion in that the sodium ion:
- has fewer electrons
  - is an isotope of sodium
  - exists only in solution
  - has a negative charge in its nucleus

**ANSWER:** a. has fewer electrons

2. An atom of strontonium-90 ( $^{90}_{38}\text{Sr}$ ) contains
- 38 electrons, 38 protons, 52 neutrons
  - 38 electrons, 38 protons, 90 neutrons
  - 52 electrons, 52 protons, 38 neutrons
  - 52 electrons, 38 protons, 38 neutrons

**ANSWER:** a. 38 electrons, 38 protons, 52 neutrons

3. How many neutrons are in ( $^{37}_{17}\text{Cl}$ ) Cl-?
- 17
  - 20
  - 21
  - 37

**ANSWER:** b. 20

4. How many protons, neutrons, and electrons are present in ( $^{47}_{22}\text{Ti}^{3+}$ ) Ti  $3^{+}$ ?
- 22p<sup>+</sup>, 25n<sup>0</sup>, 19 e<sup>-</sup>
  - 22p<sup>+</sup>, 25n<sup>0</sup>, 25e<sup>-</sup>
  - 25p<sup>+</sup>, 22n<sup>0</sup>, 19e<sup>-</sup>

# General Chemistry Acs Exam

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