

**Chem 101—General Chemistry
Practice Final Exam**

Name _____

$h = 6.626 \times 10^{-34} \text{ J s}$ (Planck's Constant)

$c = 3.00 \times 10^8 \text{ m/s}$ (speed of light)

$R_{\text{H}} = 1.097 \times 10^{-7} \text{ m}^{-1}$ (Rydberg Constant)

Multiple Choice (5 points each)

Identify the letter of the choice that best completes the statement or answers the question and legibly write the letter in the space preceding the problem number.

- _____ 1. Which of the following is not a solution?
- red wine
 - alcohol in water
 - brass
 - air
 - ice in water
- _____ 2. Which of the following statements concerning the atom is false?
- Chemical reactions involve only the electrons.
 - The atom contains protons, neutrons and electrons.
 - Protons are found in orbits about the nucleus.
 - The nucleus contains both protons and neutrons.
 - Most of the mass of the atom is found in the nucleus.
- _____ 3. Which of the following statements about two isotopes is false?
- They will have the same charge on the nucleus.
 - They will have different numbers of neutrons.
 - They will have essentially the same chemical reactivity.
 - They will have the same atomic numbers.
 - They will have the same atomic weights.
- _____ 4. Which element can be classified as an alkaline earth metal?
- Al
 - Mg
 - Ag
 - Au
 - Na
- _____ 5. Which combination below corresponds to a cation with a +1 charge?
- 7 protons and 8 electrons
 - 11 protons and 10 electrons
 - 18 protons and 18 electrons
 - 17 protons and 18 electrons
 - 20 protons and 18 electrons

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Emilie Sanchez



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