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**CHAPTER 11
INTRODUCTION TO ORGANIC CHEMISTRY**

- 11.1** Carbon can form more compounds than most other elements because carbon atoms are able to form not only single, double, and triple carbon-carbon bonds but also to link up with each other in chains and ring structures.
- 11.2** Aliphatic hydrocarbons do not contain the benzene group, or the benzene ring, whereas aromatic hydrocarbons contain one or more benzene rings.
- 11.3** Alkanes are known as saturated hydrocarbons because they contain the maximum number of hydrogen atoms that can bond with the number of carbon atoms present. Unsaturated hydrocarbons are compounds with double or triple carbon-carbon bonds. Ethane (CH_3CH_3) is a saturated hydrocarbon, and ethylene (CH_2CH_2) is an unsaturated hydrocarbon.
- 11.4** Isomers that differ in the order in which atoms are connected are called structural isomers.
- 11.5** Conformations are different spatial arrangements of a molecule that are generated by rotation about single bonds. In the staggered conformation of ethane, the three H atoms on one C atom are pointing away from the three H atoms on the other C atom, whereas in the eclipsed conformation the two groups of H atoms are aligned parallel to one another. A simpler and effective way of viewing these two conformations is using the

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