

# Microbiology Exam 2

Chapters 7, 13, 8, 20, 15

## CHAPTER 7: THE CONTROL OF MICROBIAL GROWTH

- Chemical Methods of Microbial Control
  - Disinfection and sterilization
    - Enveloped viruses are more susceptible to sterilization
  - Less reliable than heat, but is more accessible
  - Sterilants: used to sterilize instruments
  - Disinfectants
    - High level disinfectants: reduce number of microorganisms to great extent (instruments)
    - Intermediate level disinfectant: used on stethoscope and things used for outpatient testing that don't have time to be fully sterilized
    - Low level disinfectant: used every day in kitchen/bathrooms
  - How do we select a germicidal chemical?
    - Is it toxic to humans?
    - Can it withstand the presence of organic matter?
    - Is it compatible with different materials?
    - Does it leave a residue? If yes has to be rinsed with sterile water
    - Cost and availability (effective=expensive)
    - Storage and stability (most have a strict shelf life and must be diluted to be stable)
    - Environmental risk (disposal impact on water/environment)
  - Phenols (carbolic acid) and Phenolics: very effective in reducing infection risk, but very corrosive
    - Phenolic: derivative of a phenol (chemically altered to reduce irritating qualities)
      - Lysol is a phenolic (family is cresols, derived from coal tar)
    - Phenols: controls odor in sewage, irritates skin and has bad odor
  - Bisphenols
    - Derivatives of phenol that contain two phenolic groups
    - Hexachlorophane: used for degerming, prevents staph used in nurseries with caution because it causes neurological problems
  - Biguanides
    - Usually effective against gram + bacteria, and some gram - (not pseudomonas)
    - Chlorhexidine: effective on skin and mucus membrane (includes mouthwash and topical creams)
  - Halogens (antimicrobial agents)
    - Chlorine (Water treatment and gas): hypochlorous acid (disinfectant), calcium hypochlorite (cleans utensils in food industry), sodium hypochlorite (bleach), Chlorine Dioxide (used to clean water), Chloramines (chlorine and ammonium to treat water and clean)
    - Iodine (antiseptic): Betadine (combined with organic molecule to allow slow release over time), Tincture (combined with alcohol that causes pain and stains skin brown)
  - Alcohols: disinfectant and antiseptic
    - Effectively kill bacteria and fungi, not endospores/non-enveloped viruses
    - Coagulation of proteins, enzymes, and damage membranes
    - Used to enhance other chemicals (works better when combined) Isopropanol (rubbing alcohol)
  - Heavy Metal and their compounds
    - Silver (very effective for burn victims, pseudomonas), Mercury, copper, zinc (most are too toxic to use and inhibit growth of bacteria)
  - Surface Active agents/Surfactants reduce surface tension in molecules of a liquid
    - soaps, detergents
    - Acid-Anionic Sanitizers: used to clean utensils in food industry( have a negative charge that doesn't work w/ microbes)
    - Quaternary Ammonium Compounds (QUATS): used in mouthwash as a mild antiseptic
    - Cepacol, Zephiran

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**Sebastian Brünink**



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