

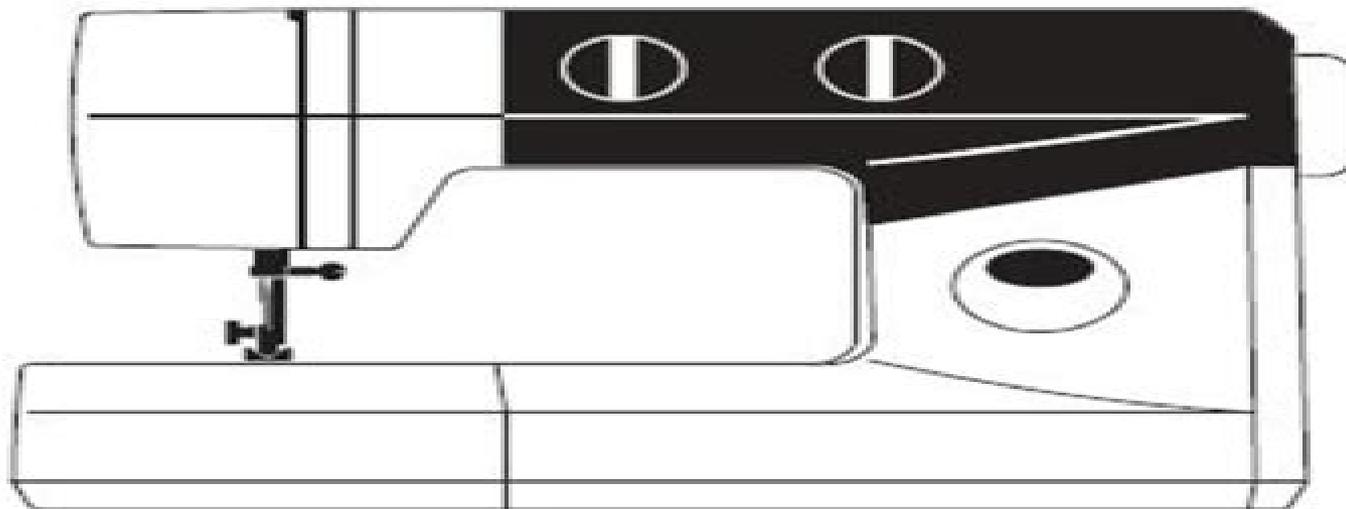
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# User's Guide

## Huskystar

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# Huskystar 207 User Guide

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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a Al b  $\text{Ba}^{2+}$  c Br<sub>2</sub> d Ca e  $\text{Ga}^{3+}$  f H<sub>2</sub> g H

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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$   $24 \text{ 3 Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3 \text{ Ag}$  oxidation reduction  $25 \text{ Cu} + \text{Pb} \rightarrow \text{Cu} + \text{Pb}^{2+}$  oxidation reduction  $26 \text{ O}_2$

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 $\text{5 c o} + \text{i}_2 \rightarrow \text{5 c o}_2$

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