



Introduction to Acids & Bases

Acids are substances that have a pH below 7, are sour to taste, and produce hydrogen ions when in solution. For example, vinegar and lemon juice are common household acids. **Bases** have a pH above 7, are bitter to taste, and produce hydroxide ions when in solution. Drain cleaner and ammonia are common household bases. A substance that is **neutral** - neither an acid or base - has a pH of 7. Pure water has a pH of 7.

Below is a table of other household items ordered by their pH values.



Note: substances that are more acidic have a lower pH and substances that are more basic have a higher pH.

Entry Level Question

Is each of the following substances an acid or base?

Sparkling water	_____	Toothpaste	_____
Orange	_____	Laundry soap	_____
Dishwater liquid	_____	Ketchup	_____
Baking soda	_____		

MORE CHALLENGING - Is each of the following substances an acid or base?

Coffee	_____
Spinach	_____
Broccoli	_____
Cayenne pepper	_____

Lecture 29 Acids Base I Worksheet

ML Yell



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Introduction to Chemistry, Laboratory Manual T. R. Dickson, 1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

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