

Answer: D. 5.0

7. Calculate the concentration of hydrogen ions in the following acetic acid solution. $pK_a = 4.76$

$$[\text{CH}_3\text{COOH}] / [\text{CH}_3\text{COO}^-] = 0.5$$

A. $7.89 \times 10^{-5} \text{M}$

B. $1.29 \times 10^{-5} \text{M}$

C. $3.39 \times 10^{-5} \text{M}$

D. $1.29 \times 10^{-4} \text{M}$

Answer: C. $3.39 \times 10^{-5} \text{M}$

8. Which molecule contains all of the following types of bonds except

A. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$

B. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$

C. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CHO}$

D. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{COOH}$

E. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{C}(\text{OH})_2\text{CH}_3$

Answer: E. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{C}(\text{OH})_2\text{CH}_3$

9. The degree of unsaturation for cyclohexene is



A. 0

B. 1

C. 2

D. 3

E. 4

Answer: C. 2

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