

CHM 1045 EXERCISE 10: -- ELECTROLYSIS --
ELECTROLYSIS

Directions: This assignment is designed to evaluate your basic understanding of basic electrochemistry concepts and provide further insight into electrochemical concepts. There is a diagram that that illustrates the electrochemical cell for this system. Use the following facts. Electrolyte cell contains an inert electrode. Part (a) is, where each correct page.

Part 1 -- Electrolytic Cell

1. Select all that apply that are correct for the cell.
2. Select all that are correct by drawing a definite circuit and voltage.
3. Select all that are correct and not over-simplified circuit.
4. All substances are made of the same and in the same.
5. All substances are made of different substances and different components.
6. All substances are made of the same and in the same.
7. All substances are made of the same and in the same.
8. All substances are made of the same and in the same.
9. All substances are made of the same and in the same.
10. All substances are made of the same and in the same.

Part 2 -- Chemical Elements and Symbols

Write the name of the element representing the following symbols. Write answer above.

1. Cu 2. H 3. Cl 4. N 5. K 6. Fe 7. Au 8. C 9. Ag 10. Pb

Part 3 -- Chemical and Physical Changes

Classify each change as either chemical or physical. Write with the name of your change (your answer) above.

1. Boiling water
2. Salt dissolved in water
3. All melting pieces of iron
4. Bread being burned in the toaster
5. Iron nails making magnetism

Introduction To Science Gradpoint Pretest Chemistry

**Peter J. Mikulecky,Michelle Rose
Gilman,Kate Brutlag**



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