

Starter unit

Vocabulary (page 4)

- 1 niece
2 aunt
3 mother
4 wife
5 sister
6 granddaughter
- 1 daughter
2 husband
3 uncle
4 nephew
5 wife
6 sister
- 1 a 3 c 5 c
2 c 4 a 6 d
- 1 grandparents
2 parents
3 children
4 uncle
5 aunt
6 nephew
7 niece
8 cousins
9 granddaughter
10 grandson
- Students' own answers.

Language focus (page 5)

- 1 1 's 4 isn't
2 're 5 aren't
3 not
- 1 We 4 It
2 They 5 He
3 I 6 you
- 1 is 5 is
2 aren't 6 is
3 isn't 7 aren't
4 are 8 isn't
- Students' own answers.
- 1 Peter's
2 parents'
3 teacher's
4 brothers'
5 Sara's
6 grandparents'
- 1 Daniel and Laura are Oliver's cousins.
2 Oliver is Emma's nephew.
3 Mark and Emma are Joseph's children.
4 Laura is David and Emma's daughter.
5 Clare is Daniel's aunt.

Vocabulary (page 6)

- 1 science 4 music
2 geography 5 English
3 PE 6 class
- 1 notes 4 teacher
2 exam 5 homework
3 book 6 room
- 1 geography 4 book
2 maths 5 class
3 exam
- Students' own answers.

Language focus (page 7)

- 1 1 Have 3 hasn't
2 haven't 4 has
- 1 haven't got
2 has got
3 haven't got
4 Has the teacher got
5 have you got
6 has she got
- 2 Has your brother got a maths exam on Friday? e has
3 Have we got two exercises for homework? c haven't
4 Has the school got a new science laboratory? d hasn't
5 Have they got an interesting geography book? b have
- 1 there aren't 5 There are
2 There are 6 there isn't
3 there are 7 there is
4 there is 8 there aren't
- Students' own answers.

Unit 1

Vocabulary (page 8)

- 1 keys 4 a key ring
2 make-up 5 jewellery
3 a purse
- 1 watch
2 money
3 bus pass
4 MP3 player
5 mobile phone
6 make-up
7 keys
8 ID card

- 1 bag 4 ticket
2 sunglasses 5 keys
3 wallet
- Students' own answers.

Language focus (page 9)

- 1 1 likes 3 doesn't
2 don't 4 like
- 1 He doesn't buy a lot of designer clothes.
2 They don't work in town.
3 Frank doesn't study a lot.
4 We don't like hip hop music.
5 She doesn't carry an ID card.
6 My dog doesn't need a drink of water.
- 1 He doesn't carry any money.
2 She watches too much TV.
3 He goes swimming twice a week.
4 We don't like horror films.
5 She doesn't wear make-up.
6 He uses an MP3 player to listen to music.
- 1 He goes
2 He doesn't play
3 She watches
4 He studies
5 She doesn't go

Vocabulary (page 10)

- 1 watch 5 meet
2 play 6 surf
3 listen to 7 collect
4 take 8 go
- 1 plays 4 take
2 collect 5 reads
3 goes 6 surf
- 1 goes 5 listen to
2 doesn't go 6 don't watch
3 reads 7 surf
4 play
- Students' own answers.

Language focus (page 11)

- 1 play 5 don't
2 Does 6 does
3 surf 7 doesn't
4 do

English Plus 2 Workbook Key

Brian Hart, Herbert Puchta, Jeff Stranks

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English Plus: 2: Workbook with Online Practice Janet Hardy-Gould, 2013-09-26 Workbook Six pages of additional practice for each Student's Book unit including exercises for vocabulary grammar reading and writing all at three levels of difficulty Workbook Two page progress review after every unit with self assessments and I can statements Workbook Reference section which includes a Language focus reference a pronunciation bank an alphabetical Wordlist with illustrations and a phonetic chart an Expression bank and an Irregular verbs list Workbook Answer Key which can be found on the English Plus Teacher's site and on iTools Online Practice Allows you to assign extra grammar vocabulary and skills based activities for the classroom or home and track your students progress Online Practice Features 250 activities per level including integrated video and audio speak and record practice and writing practice Online Practice Unlimited attempts on activities and the ability to check the correct answer inspires autonomous learning Online Practice Set writing and speaking practice that you can mark and give one to one feedback on outside the classroom to free up teaching time Online Practice Extra tools for teachers include creating groups of students for differentiated instruction and setting up online discussions on topics from the Student's Book Online Practice The access code for Online Practice is on a card with the Workbook

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Table of Contents English Plus 2 Workbook Key

1. Understanding the eBook English Plus 2 Workbook Key
 - The Rise of Digital Reading English Plus 2 Workbook Key
 - Advantages of eBooks Over Traditional Books
2. Identifying English Plus 2 Workbook Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an English Plus 2 Workbook Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from English Plus 2 Workbook Key
 - Personalized Recommendations
 - English Plus 2 Workbook Key User Reviews and Ratings
 - English Plus 2 Workbook Key and Bestseller Lists

5. Accessing English Plus 2 Workbook Key Free and Paid eBooks
 - English Plus 2 Workbook Key Public Domain eBooks
 - English Plus 2 Workbook Key eBook Subscription Services
 - English Plus 2 Workbook Key Budget-Friendly Options
6. Navigating English Plus 2 Workbook Key eBook Formats
 - ePub, PDF, MOBI, and More
 - English Plus 2 Workbook Key Compatibility with Devices
 - English Plus 2 Workbook Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of English Plus 2 Workbook Key
 - Highlighting and Note-Taking English Plus 2 Workbook Key
 - Interactive Elements English Plus 2 Workbook Key
8. Staying Engaged with English Plus 2 Workbook Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers English Plus 2 Workbook Key
9. Balancing eBooks and Physical Books English Plus 2 Workbook Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection English Plus 2 Workbook Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine English Plus 2 Workbook Key
 - Setting Reading Goals English Plus 2 Workbook Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of English Plus 2 Workbook Key
 - Fact-Checking eBook Content of English Plus 2 Workbook Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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 $2 \text{ CO}_2 \text{ (g)} + \text{H}_2\text{O (g)} \rightarrow \text{C}_2\text{H}_2 \text{ (g)} + \frac{5}{2} \text{O}_2 \text{ (g)}$ $\Delta h = 230 \text{ kJ}$
 $2 \text{ H}_2 \text{ (g)} + \text{C}_2\text{H}_6 \text{ (g)} \rightarrow \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O (g)} + \frac{1}{2} \text{O}_2 \text{ (g)} \rightarrow \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ (g)} + \frac{7}{2} \text{O}_2 \text{ (g)} \rightarrow 2 \text{ CO}_2 \text{ (g)} + 3 \text{ H}_2\text{O (g)}$ $\Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ (g)}$

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 $\text{C}_2\text{H}_4 \text{ (g)} + 3 \text{ O}_2 \text{ (g)} \rightarrow 2 \text{ CO}_2 \text{ (g)} + 2 \text{ H}_2\text{O (l)}$ $\Delta h = 1411 \text{ kJ}$ mole
 $\text{C}_2\text{H}_6 \text{ (g)} + 7 \text{ O}_2 \text{ (g)} \rightarrow 2 \text{ CO}_2 \text{ (g)} + 3 \text{ H}_2\text{O (l)}$ $\Delta h = 1560 \text{ kJ}$ mole
 $\text{H}_2 \text{ (g)} + \frac{1}{2} \text{O}_2 \text{ (g)} \rightarrow \text{H}_2\text{O (l)}$ $\Delta h = 285 \text{ kJ}$ mole

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 $\text{C}_2\text{H}_4 \text{ (g)} + 2 \text{ NO}_2 \text{ (g)} \rightarrow \text{C}_2\text{H}_6 \text{ (g)} + 2 \text{ O}_2 \text{ (g)} + \text{N}_2\text{O}_5 \text{ (g)}$
 given the following information

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web libretxts states hess s law states that regardless of the multiple stages or steps of a reaction the total enthalpy change for the reaction is the sum of all changes libretxts quot hess s law quot 2019 which determines the enthalpy change value hess s law is written as $\Delta H_{\text{target}} = \Delta H_1 + \Delta H_2 + \Delta H_3$ or $\Delta H_{\text{target}} = \sum \Delta H_{\text{known}}$

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web hess s law states that if a process can be expressed as the sum of two or more steps the enthalpy change for the overall process is the sum of the ΔH values for each step to use hess s law two principles must be understood one if an equation is reversed the sign of the ΔH value is also reversed two if an equation is multiplied by a

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web hess s law ws 6a answer key directions for each of the following problems find the unknown ΔH and show the reactions adding up to the overall reaction

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web hess s law labs by austin lee alayna baron lily zmachinski introduction in order to calculate the enthalpy change for the combustion of magnesium oxide $\text{Mg} + \frac{1}{2}\text{O}_2 \rightarrow \text{MgO}$ we used

additivity of heats of reaction hess s law lab report - Aug 03 2022

web hess s law states that the total enthalpy change for the reaction will be the sum of all those changes no matter how many different steps or stages in the reaction there are cohen 2016 the equations for the reactions in the experiment done are as follows $\text{NaOH} + \text{Na} \rightarrow \text{NaOH} + \text{Na}$

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web hess s laws abstract the following experiment was designed using hess s law and simple calorimetry to determine the enthalpy for NaOH hess s law says that the overall change in enthalpy for the reaction is the sum of all changes regardless of the different stages or phases of a reaction

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web hess s law states that if a reaction is the sum of two or more other reactions the ΔH for the overall process must be the sum of the ΔH values of the constituent reactions enthalpy change ΔH is independent of the path that a reaction follows to move from reactants

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web this activity provides a demonstration of hess law using three reactions the solubility NaOH in water the solubility NaOH in HCl and the reaction of a solution of HCl and a solution of NaOH

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web hess s law worksheet answers 1 calculate ΔH for the reaction $C_2H_4(g) + H_2(g) \rightarrow C_2H_6(g)$ from the following data $C_2H_4(g) + 3O_2(g) \rightarrow 2CO_2(g) + 2H_2O(l) \Delta H = 1411 \text{ kJ}$ $C_2H_6(g) + \frac{3}{2}O_2(g) \rightarrow 2CO_2(g) + 3H_2O(l) \Delta H = 1560 \text{ kJ}$

7a first law enthalpy calorimetry and hess s law worksheet - Oct 05 2022

web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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web may 8 2015 hess law lab by maya parks partners ben seufert kelsea floyd convert joules to kJ in your final answer multiply the mass by the change in temperature and the c given trial 1 use hess s law and the following equations and ΔH values to determine the heat of reaction for the reaction C_2H_4

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web thus for reaction 9 2 in which 1 mole of H_3O^+ and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta H = 55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}}$

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web hess s law enthalpy of reaction and hess s law abstract the goal of this lab is to perform experiments and practice hess s law by determining enthalpy change in chemical reactions in this lab three experiments were conducted in order to test hess s law the experiments required observing and collecting data on three different reactions

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web objective the purpose of the lab is to combine two reactants in the calorimeter and measure the heats of reaction in solution eventually allowing an individual to prove or disprove hess s law introduction thermodynamics is the study of energy exchanged between the system and the surroundings

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions $1. NaOH(s) + NaCl(aq) \rightarrow H_2O(l) + 2. NaOH(s) + HCl(aq) \rightarrow H_2O(l) + 3. NaCl(aq) + H_2O(l)$