

- (c) 350 cm, convergent  
(d) 150 cm, divergent
16. Three resistances A, B and C each of  $5\Omega$  are connected to a battery of 10 V as shown in the figure. Calculate the current through C:
- 
- (a) 2 amp  
(b) 1 amp  
(c) 3 amp  
(d) 0.6 amp
17. The device used to generate electrical energy is:
- (a) Electric motor  
(b) Generator  
(c) Galvanometer  
(d) Voltmeter
18. The number of molecules in 11 g of  $\text{CO}_2$  are:
- (a)  $0.25 \times 10^{23}$   
(b)  $0.50 \times 10^{23}$   
(c)  $1.00 \times 10^{23}$   
(d)  $1.51 \times 10^{23}$
19. The value of charge / mass ratio of electron was determined by:
- (a) W.K. Roentgen  
(b) J.J. Thomson  
(c) Marie Curie  
(d) Niels Bohr
20. Which of the following reactions is an example of combination reaction?
- (a)  $\text{Zn}(s) + \text{CuSO}_4(aq) \rightarrow \text{ZnSO}_4(aq) + \text{Cu}(s)$   
(b)  $\text{BaCl}_2(aq) + \text{Na}_2\text{SO}_4(aq) \rightarrow \text{BaSO}_4(s) + 2\text{NaCl}(aq)$   
(c)  $\text{CaCO}_3(s) \rightarrow \text{CaO}(s) + \text{CO}_2(g)$   
(d)  $\text{C}(s) + \text{O}_2(g) \rightarrow \text{CO}_2(g)$
21. The element 'X' and 'Y' have atomic numbers 12 and 17 respectively. Element 'X' reacts with element 'Y' to form compound with molecular formula:
- (a) XY  
(b)  $\text{XY}_2$   
(c)  $\text{XY}_3$   
(d)  $\text{X}_3\text{Y}_2$
22. The atomic radius (pm) of Li, Na, K and Rb varies in the order:
- (a)  $\text{Na} < \text{K} < \text{Rb} < \text{Li}$   
(b)  $\text{K} < \text{Na} < \text{Li} < \text{Rb}$   
(c)  $\text{Li} < \text{Na} < \text{K} < \text{Rb}$   
(d)  $\text{Rb} < \text{K} < \text{Na} < \text{Li}$
23. The electron affinity value ( $\text{kJ mol}^{-1}$ ) of fluorine (F) is less than:
- (a) Hydrogen (H)  
(b) Lithium (Li)  
(c) Oxygen (O)  
(d) Chlorine (Cl)
24. Which of the following is an example of strong electrolyte?
- (a)  $\text{H}_2\text{CO}_3$   
(b)  $\text{NH}_4\text{OH}$   
(c) NaCl  
(d)  $\text{COOH} - \text{COOH}$
25. If pH of a solution changes from 5 to 4, the change in hydrogen ion concentration shall be:
- (a) two times  
(b) five times  
(c) ten times  
(d) twenty times
26. Bleaching powder is manufactured by the reaction of:
- (a)  $\text{CaCl}_2$  and  $\text{CaCO}_3$   
(b)  $\text{Cl}_2$  and  $\text{Ca}(\text{OH})_2$   
(c)  $\text{Cl}_2$  and  $\text{CaSO}_4$   
(d)  $\text{Cl}_2$  and  $\text{Ca}(\text{HCO}_3)_2$
- The hard glass is obtained by fusing:
- (a) soda ash, sand and limestone  
(b) a mixture of sand, lime, borax and alkali carbonates  
(c) potassium carbonate and limestone  
(d) potassium carbonate, lead oxide and sand
27. Which of the following metals is most reactive?
- (a) Aluminium  
(b) lead  
(c) Mercury  
(d) Silver
28. Which of the following alloys contains chromium?
- (a) Steel  
(b) Stainless steel  
(c) Magnalium  
(d) Brass
29. Which of the following is monomer of natural rubber?
- (a) Chloroethene  
(b) Chloroprene  
(c) Isoprene  
(d) Buta-1, 3-diene
30. Alkaline  $\text{KMnO}_4$  oxidises propanone to:
- (a) propanoic acid  
(b) ethanoic acid  
(c) methanoic acid  
(d) oxalic acid
31. Which of the following metals can displace Zn from  $\text{ZnSO}_4$  solution?
- (a) Calcium  
(b) Copper  
(c) Iron  
(d) Mercury
32. Heating of sodium ethanoate with soda lime yields:
- (a) ethane  
(b) methane  
(c) ethanol  
(d) methanol
33. The compound formed by the reaction of ethyne with bromine is:
- (a)  $\text{Br}-\text{CH}=\text{CH}-\text{Br}$   
(b)  $\text{Br}-\text{CH}_2-\text{CHBr}_2$   
(c)  $\text{CH}_2=\text{CH}-\text{Br}$   
(d)  $\text{Br}_3\text{CH}-\text{CHBr}_2$
34. Cell were first discovered by Robert Hooke in the year:
- (a) 1665  
(b) 1674  
(c) 1831  
(d) 1839
35. The plant cells have a rigid cell wall that lies:
- (a) Outside the plasma membrane  
(b) Inside the plasma membrane  
(c) In between the plasma membranes  
(d) None of the above
36. During mitosis nucleolus and nuclear membrane are lost in stage:
- (a) Prophase  
(b) Metaphase  
(c) Anaphase  
(d) Telophase
37. An example of simple and permanent tissue is:
- (a) Xylem  
(b) Phloem  
(c) Sclerenchyma  
(d) All the above
38. The shape of squamous epithelial tissue is:
- (a) Cubical  
(b) Flattened  
(c) Pillar like  
(d) None of the above
39. Fresh water sponge belongs to phylum:
- (a) Arthropoda  
(b) Annelida

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1. Understanding the eBook Entrance Exam Question Papers For Class 11
  - The Rise of Digital Reading Entrance Exam Question Papers For Class 11
  - Advantages of eBooks Over Traditional Books
2. Identifying Entrance Exam Question Papers For Class 11
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Entrance Exam Question Papers For Class 11
  - User-Friendly Interface
4. Exploring eBook Recommendations from Entrance Exam Question Papers For Class 11
  - Personalized Recommendations
  - Entrance Exam Question Papers For Class 11 User Reviews and Ratings
  - Entrance Exam Question Papers For Class 11 and Bestseller Lists
5. Accessing Entrance Exam Question Papers For Class 11 Free and Paid eBooks
  - Entrance Exam Question Papers For Class 11 Public Domain eBooks
  - Entrance Exam Question Papers For Class 11 eBook Subscription Services
  - Entrance Exam Question Papers For Class 11 Budget-Friendly Options
6. Navigating Entrance Exam Question Papers For Class 11 eBook Formats

- ePub, PDF, MOBI, and More
  - Entrance Exam Question Papers For Class 11 Compatibility with Devices
  - Entrance Exam Question Papers For Class 11 Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Entrance Exam Question Papers For Class 11
    - Highlighting and Note-Taking Entrance Exam Question Papers For Class 11
    - Interactive Elements Entrance Exam Question Papers For Class 11
  8. Staying Engaged with Entrance Exam Question Papers For Class 11
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Entrance Exam Question Papers For Class 11
  9. Balancing eBooks and Physical Books Entrance Exam Question Papers For Class 11
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Entrance Exam Question Papers For Class 11
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Entrance Exam Question Papers For Class 11
    - Setting Reading Goals Entrance Exam Question Papers For Class 11
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Entrance Exam Question Papers For Class 11
    - Fact-Checking eBook Content of Entrance Exam Question Papers For Class 11
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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