

**Q3)** Attempt any four of the following :

- a) Prove that slope of adiabatic curve through a point in P-V diagram is  $\gamma$  times the slope of the isothermal curve through the same point. [4]
- b) Show that the entropy remains constant during a reversible cyclic change. [4]
- c) What is Boyle's temperature? Obtain the relation between Boyle temperature and critical temperature. [4]
- d) Find the temperature in Fahrenheit scale, Kelvin scale and Reaumur scale corresponding to  $45^{\circ}\text{C}$ . [4]
- e) The expansion ratio and compression ratio are 6 and 12 respectively. If the value of  $\gamma$  is 1.4 for the working substance in a Diesel engine. Find its efficiency. [4]
- f) A reversible engine converts  $1/8^{\text{th}}$  of heat into work. When temperature of the sink is decreased by  $62^{\circ}\text{K}$ , its efficiency is doubled. Find the temperature of the source. [4]

**Q4)** Attempt any two of the following :

- a) State the working principle of venturimeter and discuss in detail the working of venturimeter. [8]
- b) i) A position of particle along x - axis with time is given by  $x(t) = 7 + 5t + 8t^2$  where x in meter.
  - 1) Find the average velocity between  $t = 0$  and  $t = 1$  sec. [4]
  - 2) Find the value of acceleration at  $t = 0.5$  and  $t = 1$  sec. [4]
- ii) State the principle of conservation of energy and show that the total mechanical energy remain constant. [4]
- c) i) Calculate the work done in moving a particle along a vector  $\vec{r} = (3\vec{i} - \vec{j} + 6\vec{k})$  meter, if the applied force is  $\vec{F} = (\vec{i} + 3\vec{j} + 2\vec{k})$  Newton. [4]
- ii) Obtain the equation of continuity. [4]

**Q5)** Attempt any two of the following :

- a) Describe Carnot's heat engine. Obtain an expression for its efficiency. [8]
- b) i) Calculate the work done when a gram molecule of a gas expands isothermally at  $37^{\circ}\text{C}$  to double its original volume ( $R = 8.3$  joule/degree/mole). [4]
- ii) Derive reduced equation of state from van-der-waals equation of state. [4]
- c) i) Calculate the van-der-waal's constant 'a' and 'b' for dry air given that  $T_c = 132^{\circ}\text{K}$ ,  $P_c = 37.2$  atmp. and  $R = 82.07$   $\text{cm}^3\cdot\text{atmp}/^{\circ}\text{K}$ . [4]
- ii) Explain construction and working of liquid filled thermometer. [4]



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