

- e) Find the wave number of second line of Paschen series. (Given: $R = 1.097 \times 10^7 \text{ m}^{-1}$)
- f) A microwave radiation has frequency of 14 GHz. What would be the wavelength and energy of photon corresponding to this radiation. (Given: $h = 6.626 \times 10^{-34} \text{ Js}$)

Q3) Attempt any four of the following:

[16]

- a) Using Biot-Savart's law, obtain an expression for magnetic field produced in long straight conductor.
- b) Distinguish between diamagnetic and ferromagnetic material.
- c) Using Gauss's theorem, obtain an expression for electric intensity at a point due to uniformly charged non-conducting sphere.
- d) Calculate the electric field due to dipole of dipole moment $2 \times 10^{-30} \text{ C-m}$ at a distance of 1 m from it on its axis. (Given: $P = 2 \times 10^{-30} \text{ C-m}$, $r = 1 \text{ m}$)
- e) A bar magnet made of iron has magnetic moment 4.0 A/m^2 and mass $5 \times 10^{-3} \text{ kg}$. If the density of iron is $6 \times 10^3 \text{ kg/m}^3$, find the intensity of magnetization.
- f) A solenoid of 500 turns/m is carrying current 3A. If the core is made of iron which has a relative permeability of 5000, determine the magnitude of magnetic intensity and magnetization.

Q4) Attempt any two of the following:

[16]

- a) Explain the various sources of electromagnetic waves.
- b) i) Explain induced absorption and spontaneous emission in detail.
ii) CO molecule absorbs infrared radiations of frequency $6.42 \times 10^{13} \text{ Hz}$. What is the force constant of the bond in CO molecule and what is the spacing between the vibrational energy levels. (Given: $\mu = 1.14 \cdot 10^{-26} \text{ kg}$)
- c) i) What do you mean by covalent bond? Explain the properties of covalent compounds.
ii) The first line of Balmer series of hydrogen atom has wavelength of 6563 \AA . Calculate the wavelength of second line of Balmer series.

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Table of Contents Fybsc Question Papers 2013

1. Understanding the eBook Fybsc Question Papers 2013
 - The Rise of Digital Reading Fybsc Question Papers 2013
 - Advantages of eBooks Over Traditional Books
2. Identifying Fybsc Question Papers 2013
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Fybsc Question Papers 2013
 - User-Friendly Interface
4. Exploring eBook Recommendations from Fybsc Question Papers 2013
 - Personalized Recommendations
 - Fybsc Question Papers 2013 User Reviews and Ratings
 - Fybsc Question Papers 2013 and Bestseller Lists
5. Accessing Fybsc Question Papers 2013 Free and Paid eBooks
 - Fybsc Question Papers 2013 Public Domain eBooks
 - Fybsc Question Papers 2013 eBook Subscription Services
 - Fybsc Question Papers 2013 Budget-Friendly Options
6. Navigating Fybsc Question Papers 2013 eBook Formats

- ePub, PDF, MOBI, and More
 - Fybsc Question Papers 2013 Compatibility with Devices
 - Fybsc Question Papers 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Fybsc Question Papers 2013
 - Highlighting and Note-Taking Fybsc Question Papers 2013
 - Interactive Elements Fybsc Question Papers 2013
 8. Staying Engaged with Fybsc Question Papers 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Fybsc Question Papers 2013
 9. Balancing eBooks and Physical Books Fybsc Question Papers 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Fybsc Question Papers 2013
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Fybsc Question Papers 2013
 - Setting Reading Goals Fybsc Question Papers 2013
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Fybsc Question Papers 2013
 - Fact-Checking eBook Content of Fybsc Question Papers 2013
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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