

Unit 7 Gas Laws Test Review - Answer Key

1. A 0.4 L sample of gas is collected at a temperature of 45°C. At what temperature will the gas fill a 0.725 L container?

LIST GIVENS	FORMULA	GIVENS PLUGGED INTO FORMULA:
$P_1 =$ $V_1 = 0.4 \text{ L}$ $T_1 = 318 \text{ K}$ $P_2 =$ $V_2 = 0.725 \text{ L}$ $T_2 = ?$	$\frac{V_1}{T_1} = \frac{V_2}{T_2}$	$\frac{0.4}{318} = \frac{.725}{T_2}$ $(.725 \times 318) / 0.4 = T_2$
answer with correct units: $T_2 = 576.38 \text{ K}$		

2. If 3 moles of a gas at a pressure of 3.5 atm has a volume of 110 L, what is the temperature?

LIST GIVENS	FORMULA	GIVENS PLUGGED INTO FORMULA:
$P = 3.5 \text{ atm}$ $V = 110 \text{ L}$ $n = 3 \text{ mol}$ $R = 0.0821$ $T = ?$	$PV = nRT$	$(3.5 \times 110) = (3 \times 0.0821 \times T)$ $385 / .2463 = T$
answer with correct units: $T = 1563.13 \text{ K}$		

3. Identify if the following would cause an INCREASE or DECREASE in pressure in a gaseous system.

- Increase in the size of the container = **pressure decreases**
- More of the same gas is added to the container = **pressure increases**
- Increase in temperature = **pressure increases**
- A different gas is added to the container = **pressure increases**

4. A sealed container of Oxygen gas has a volume of 356 L. If heated from a temperature of 293-K to 318-K, what will the pressure be if the original pressure was 1.2 atm?

LIST GIVENS	FORMULA	GIVENS PLUGGED INTO FORMULA:
$P_1 = 1.2 \text{ atm}$ $V_1 =$ $T_1 = 293 \text{ K}$ $P_2 = ?$ $V_2 =$ $T_2 = 318 \text{ K}$	$\frac{P_1}{T_1} = \frac{P_2}{T_2}$	$\frac{1.2}{293} = \frac{P_2}{318}$ $(1.2 \times 318) / 293 = P_2$

Gas Laws Test Review Answers

Christian Drosten



Gas Laws Test Review Answers:

Decoding **Gas Laws Test Review Answers**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Gas Laws Test Review Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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