



## Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

## Assessment Statements

- |                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                        |                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                   |
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| <ul style="list-style-type: none"> <li><input type="checkbox"/> <b>Outline</b> the Scientific Method</li> <li><input type="checkbox"/> <b>Identify</b> independent, dependent &amp; controlled variables.</li> <li><input type="checkbox"/> <b>Demonstrate</b> correct and safe use of laboratory equipment.</li> <li><input type="checkbox"/> <b>Distinguish</b> between quantitative and qualitative data             <ul style="list-style-type: none"> <li>&gt; <i>Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.</i></li> <li>&gt; <i>Quantitative is mathematical, including</i> <ul style="list-style-type: none"> <li>↳ <i>Raw data (as recorded)</i></li> <li>↳ <i>Processed (after calculations)</i></li> </ul> </li> </ul> </li> <li><input type="checkbox"/> <b>Distinguish</b> between accuracy and precision             <ul style="list-style-type: none"> <li>&gt; <i>Accuracy describes the correctness of a set of measurements</i></li> <li>&gt; <i>Precision describes the repeatability of a set of measurements</i></li> </ul> </li> </ul> | <ul style="list-style-type: none"> <li><input type="checkbox"/> <b>State</b> SI units of measurement for             <ul style="list-style-type: none"> <li>&gt; <i>Mass: grams (g)</i></li> <li>&gt; <i>Length: Metres (m), millimeters (mm)</i></li> <li>&gt; <i>Concentration: molarity (M)</i></li> <li>&gt; <i>Temperature: Degrees Celsius (°C)</i></li> <li>&gt; <i>Volume: Litres (L) or millilitres (mL)</i></li> <li>&gt; <i>Density: grams per cubic centimeter (g/cm<sup>3</sup>)</i></li> <li>&gt; <i>Conductivity: Siemens (S)</i></li> <li>&gt; <i>Acidity or alkalinity (pH)</i></li> </ul> </li> <li><input type="checkbox"/> <b>Energy:</b> Joules (J or kJ) <b>Determine</b> the uncertainty of digital and analog measuring tools             <ul style="list-style-type: none"> <li>&gt; <i>Usually ± 1 of the smallest division</i></li> </ul> </li> <li><input type="checkbox"/> <b>Calculate</b> the % error of a given or recorded set of values             <ul style="list-style-type: none"> <li>&gt; <i><math>(\text{Measured} - \text{True}) / \text{True} \times 100</math></i></li> </ul> </li> <li><input type="checkbox"/> <b>State</b> values to appropriate numbers of significant figures</li> <li><input type="checkbox"/> <b>Calculate</b> values to appropriate numbers of significant figures</li> </ul> |
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## Related Concepts:

- 01.1 Element symbols (Quia quiz: <http://www.quia.com/quiz/3246640.htm>)
- 01.2 Accuracy, Precision, Error, Uncertainty (Quia quiz: <http://www.quia.com/quiz/3507626.htm>)
- 01.3 SigDigs and Notation (Quia quiz: <http://www.quia.com/quiz/3504378.htm>)

## Command Terms: Be sure you understand what is being asked of you

**Analyse** Break down to identify key parts. Interpret information to reach conclusions.

<b>Apply</b>	Use knowledge & understanding in response to a real situation or given circumstances.
<b>Calculate</b>	Obtain a numerical answer, showing relevant stages in the working.
<b>Define</b>	Give the precise meaning of a word, phrase, concept or physical quantity.
<b>Describe</b>	Give a detailed account or picture of an event, pattern, process or pattern.
<b>Design</b>	Produce a plan, simulation or model.
<b>Determine</b>	Obtain the only possible answer.
<b>Discuss</b>	Offer a considered and balanced review that includes a range of arguments, factors or hypotheses. Opinions or conclusions should be presented clearly and supported by appropriate evidence.
<b>Distinguish</b>	Make clear the differences between two or more concepts or items.
<b>Evaluate</b>	Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria.
<b>Explain</b>	Give a detailed account including reasons or causes.
<b>Identify</b>	Provide an answers from a number of possibilities. Recognise and state briefly a distinguishing fact or feature.
<b>Label</b>	Add title, labels or brief explanation(s) to a diagram or graph.
<b>List</b>	Give a sequence of answers with no explanation.
<b>Outline</b>	Give a brief account.
<b>Recall</b>	Remember or recognize from prior learning experiences.
<b>State</b>	Give a specific name, value or other brief answer without explanation or calculation.
<b>Suggest/Propose</b>	Propose a solution, hypothesis or other possible answer.

# Introduction To Chemistry Lab Manual Answers

**Julia Schneider**



## **Introduction To Chemistry Lab Manual Answers:**

## Whispering the Techniques of Language: An Emotional Journey through **Introduction To Chemistry Lab Manual Answers**

In a digitally-driven world wherever monitors reign supreme and instant connection drowns out the subtleties of language, the profound techniques and mental nuances concealed within phrases usually get unheard. Yet, located within the pages of **Introduction To Chemistry Lab Manual Answers** a interesting fictional treasure pulsing with fresh emotions, lies an extraordinary journey waiting to be undertaken. Composed by an experienced wordsmith, this charming opus attracts viewers on an introspective trip, delicately unraveling the veiled truths and profound affect resonating within the material of each word. Within the psychological depths of the touching review, we can embark upon a honest exploration of the book is key themes, dissect its interesting writing style, and yield to the effective resonance it evokes heavy within the recesses of readers hearts.

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