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Chapter 4: The Physical Properties of Water

Chapter 4: Quick Review Questions

4.1 The Water Molecule

1. The water molecule is asymmetrical. It consists of one oxygen atom and two hydrogen atoms. The angle between the hydrogen atoms is about 105° .
2. It is called a polar molecule because the end of the molecule with the hydrogen atoms has a positive charge and the opposite end has a balancing negative charge.
3. The shape of the molecule makes it a polar molecule. Consequently, water molecules are able to bond with other water molecules through hydrogen bonding.

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