

Chemistry Practice: Writing Chemical Formulas

Write a chemical formula for each substance.

- | | | | |
|--|--------------------------------|--|---------------------------|
| 1. <u>NaCl</u> | sodium chloride | 39. <u>Ca(NO₃)₂</u> | calcium nitrate |
| 2. <u>N₂O₅</u> | dinitrogen pentoxide | 40. <u>SO₃</u> | sulfur trioxide |
| 3. <u>H₂S</u> | hydrosulfuric acid | 41. <u>KCN</u> | potassium cyanide |
| 4. <u>K₂SO₄</u> | potassium sulfate | 42. <u>Pb(NO₃)₂</u> | lead(II) nitrate |
| 5. <u>H₂C₂O₄</u> | oxalic acid | 43. <u>H₂S</u> | hydrogen sulfide |
| 6. <u>Ag₂C₂H₃O₂</u> | silver acetate | 44. <u>CoCl₃</u> | cobalt(III) chloride |
| 7. <u>Cr(NO₃)₃</u> | chromium(III) nitrate | 45. <u>SF₆</u> | sulfur hexafluoride |
| 8. <u>HC₂H₃O₂</u> | acetic acid | 46. <u>Ca₃N₂</u> | calcium nitride |
| 9. <u>(NH₄)₂CO₃</u> | ammonium carbonate | 47. <u>CuI</u> | copper(I) iodide |
| 10. <u>Ca(OH)₂</u> | calcium hydroxide | 48. <u>SiO₂</u> | silicon dioxide |
| 11. <u>H₂C₄H₄O₆</u> | tartaric acid | 49. <u>Sn(C₂H₃O₂)₄</u> | tin(IV) acetate |
| 12. <u>Hg(NO₃)₂</u> | mercury(II) nitrate | 50. <u>CCl₄</u> | carbon tetrachloride |
| 13. <u>N₂O</u> | dinitrogen monoxide | 51. <u>CuS</u> | copper(II) sulfide |
| 14. <u>Fe₂O₃</u> | iron(III) oxide | 52. <u>Pb₃(PO₄)₂</u> | lead(II) phosphate |
| 15. <u>Pb(SiO₃)₂</u> | lead(II) chlorate | 53. <u>XeCl₄</u> | xenon tetrachloride |
| 16. <u>(NH₄)₃PO₄</u> | ammonium phosphate | 54. <u>Rb₂O</u> | rubidium oxide |
| 17. <u>ZnCl₂</u> | zinc chloride | 55. <u>MgSe</u> | magnesium selenide |
| 18. <u>Ca₃(PO₄)₂</u> | calcium phosphate | 56. <u>NH₄Cl</u> | ammonium chloride |
| 19. <u>OF₂</u> | oxygen difluoride | 57. <u>Fe(C₂H₃O₂)₃</u> | iron(III) acetate |
| 20. <u>NaN₃</u> | sodium azide | 58. <u>K₂Cr₂O₇</u> | potassium dichromate |
| 21. <u>Fe₂(SO₄)₃</u> | iron(III) sulfate | 59. <u>PBr₃</u> | phosphorous tribromide |
| 22. <u>H₃AsO₃</u> | arsenous acid | 60. <u>Na₃PO₃</u> | sodium phosphite |
| 23. <u>Cr₂O₃</u> | chromium(III) oxide | 61. <u>Na₃PO₄</u> | sodium phosphate |
| 24. <u>N₂O₄</u> | dinitrogen tetroxide | 62. <u>Hg(NO₃)₂</u> | mercury(II) nitrate |
| 25. <u>NH₄NO₃</u> | ammonium nitrate | 63. <u>LiHCO₃</u> | lithium bicarbonate |
| 26. <u>AuBr₃</u> | gold(III) bromide | 64. <u>CrF₃</u> | chromium(III) fluoride |
| 27. <u>CO</u> | carbon monoxide | 65. <u>PbI₂</u> | lead(II) iodide |
| 28. <u>K₂CO₃</u> | potassium carbonate | 66. <u>H₂SO₃</u> | sulfurous acid |
| 29. <u>HIo₃</u> | iodic acid | 67. <u>SnF₂</u> | tin(II) fluoride |
| 30. <u>CsCl</u> | cesium chloride | 68. <u>HgCrO₄</u> | mercury(II) chromate |
| 31. <u>Ni(MnO₄)₂</u> | nickel(II) permanganate | 69. <u>KNO₃</u> | potassium nitrate |
| 32. <u>Al₂(SO₄)₃</u> | aluminum sulfate | 70. <u>SrCl₂</u> | strontium chloride |
| 33. <u>Al₂(SO₃)₃</u> | aluminum sulfite | 71. <u>P₄O₁₀</u> | tetraphosphorous decoxide |
| 34. <u>Ba(C₂H₃O₂)₂</u> | barium acetate | 72. <u>KNO₃</u> | potassium nitrate |
| 35. <u>Mn(OH)₃</u> | manganese(III) hydroxide | 73. <u>KNO₂</u> | potassium nitrite |
| 36. <u>KH₂PO₄</u> | potassium dihydrogen phosphate | 74. <u>K₃N</u> | potassium nitride |
| 37. <u>HF</u> | hydrofluoric acid | 75. <u>CaO</u> | calcium oxide |
| 38. <u>Al(BrO₃)₃</u> | aluminum bromate | 76. <u>Fe(IO₄)₂</u> | iron(II) periodate |

Lab 12 Chemical Names And Formulas Answers

JL Elias



Lab 12 Chemical Names And Formulas Answers:

Lab 12 Chemical Names And Formulas Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Lab 12 Chemical Names And Formulas Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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