

Reteaching 8-3

Multiplying Binomials

OBJECTIVE: Multiplying binomials	MATERIALS: None
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To multiply two binomials, follow these steps:

- Multiply each term in one binomial by each term of the other binomial. Drawing arrows on a visual reminder of what to do is a helpful technique.
- Circle like terms and combine.

Example

Find the product $(x + 7)(x + 2)$.

$$(x + 7)(x + 2)$$

— Draw arrows from the first term in the first binomial to both terms in the second binomial.

$$x^2 + 2x$$

— Multiply each term of the second binomial by x .

$$(x + 7)(x + 2)$$

— Draw arrows from the second term in the first binomial to both terms in the second binomial.

$$7x + 14$$

— Multiply each term of the second binomial by 7.

$$x^2 + 2x + 7x + 14$$

— Add the two expansions.

$$x^2 + \textcircled{2x} + \textcircled{7x} + 14$$

— Circle like terms and combine.

$$x^2 + 9x + 14$$

— Solution

Exercises

Use arrows as shown above to simplify each product.

1. $(x + 6)(x + 2)$

2. $(x + 8)(x + 4)$

3. $(x + 3)(x + 9)$

4. $(x + 2)(x + 7)$

5. $(2x + 3)(x + 4)$

6. $(x + 4)(2x + 5)$

Simplify each product.

7. $(7x + 4)(2x + 4)$

8. $(3x + 2)(5x + 2)$

9. $(5x + 3)(x + 1)$

10. $(2x + 1)(x + 1)$

11. $(4x + 1)(2x + 1)$

12. $(3x + 1)(x + 2)$

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Lesson 8 Reteaching With Practice Answers Key

R Bogdan



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Glencoe Writer's Choice, 2005 **Scott Foresman-Addison Wesley Mathematics**, 2008 **El-Hi Textbooks & Serials in Print** Bowker Editorial Staff, 2004-02 **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Teacher's Wraparound Edition: Two Biology Everyday Experience* Albert Kaskel, 1994-04-19 **Invitation to Mathematics: Teacher's edition**, 1985 *Children's Books in Print* R R Bowker Publishing, Bowker, 1999-12 **Teaching Behavior** Terrance M. Scott, 2016-05-26 The key to effective classroom management starts with instruction Teaching Behavior goes beyond setting classroom rules communicating consequences and providing the usual tips on engaging students and building relationships It draws on the most current evidence based practices at the heart of effective teaching so you can maximize student success Ideal as a teacher guide or textbook it offers New insights on why instruction is the foundation for all student behavior Practical tools for managing all types of students and classrooms including the most challenging Self assessment checklists and discussion questions for teacher book study groups Accompanying video modules for each chapter *High Leverage Practices for Inclusive Classrooms* James McLeskey, Lawrence Maheady, Bonnie Billingsley, Mary T. Brownell, Timothy J. Lewis, 2022-03-30 High Leverage Practices for Inclusive Classrooms Second Edition offers a set of practices that are integral to the support of student learning and that can be systematically taught learned and implemented by those entering the teaching profession In this second edition chapters have been fully updated to reflect changes in the field since its original publication and feature all new examples illustrating the use of HLPs and incorporating culturally responsive practices Focused primarily on Tiers 1 and 2 or work that mostly occurs with students with mild to moderate disabilities in general education classrooms this powerful research based resource provides rich practical information highly suitable for teachers and additionally useful for teacher educators and teacher preparation programs

Teaching and Learning STEM Richard M. Felder, Rebecca Brent, 2024-03-13 The widely used STEM education book updated Teaching and Learning STEM A Practical Guide covers teaching and learning issues unique to teaching in the science technology engineering and math STEM disciplines Secondary and postsecondary instructors in STEM areas need to master specific skills such as teaching problem solving which are not regularly addressed in other teaching and learning books This book fills the gap addressing topics like learning objectives course design choosing a text effective instruction active learning teaching with technology and assessment all from a STEM perspective You ll also gain the knowledge to implement learner centered instruction which has been shown to improve learning outcomes across disciplines For this edition chapters have been updated to reflect recent cognitive science and empirical educational research findings that inform STEM pedagogy You ll also find a new section on actively engaging students in synchronous and asynchronous online courses and content has been substantially revised to reflect recent developments in instructional technology and online course development and delivery Plan and deliver lessons that actively engage students in person or online Assess students

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[Get Better Faster](#) Paul Bambrick-Santoyo,2016-07-25 Effective and practical coaching strategies for new educators plus valuable online coaching tools Many teachers are only observed one or two times per year on average and even among those who are observed scarcely any are given feedback as to how they could improve The bottom line is clear teachers do not need to be evaluated so much as they need to be developed and coached In [Get Better Faster A 90 Day Plan for Coaching New Teachers](#) Paul Bambrick Santoyo shares instructive tools of how school leaders can effectively guide new teachers to success Over the course of the book he breaks down the most critical actions leaders and teachers must take to achieve exemplary results Designed for coaches as well as beginning teachers [Get Better Faster](#) is an integral coaching tool for any school leader eager to help their teachers succeed [Get Better Faster](#) focuses on what s practical and actionable which makes the book s approach to coaching so effective By practicing the concrete actions and micro skills listed in [Get Better Faster](#) teachers will markedly improve their ability to lead a class producing a steady chain reaction of future teaching success Though focused heavily on the first 90 days of teacher development it s possible to implement this work at any time Junior and experienced teachers alike can benefit from the guidance of [Get Better Faster](#) while at the same time closing existing instructional gaps Featuring valuable and practical online training tools available at <http://www.wiley.com/go/getbetterfaster> [Get Better Faster](#) provides agendas presentation slides a coach s guide handouts planning templates and 35 video clips of real teachers at work to help other educators apply the lessons learned in their own classrooms [Get Better Faster](#) will teach you The core principles of coaching Go Granular Plan Practice Follow Up Repeat Make Feedback More Frequent Top action steps to launch a teacher s development in an easy to read scope and sequence guide It also walks you through the four phases of skill building Phase 1 Pre Teaching Dress Rehearsal Phase 2 Instant Immersion Phase 3 Getting into Gear Phase 4 The Power of Discourse Perfect for new educators and those who supervise them [Get Better Faster](#) will also earn a place in the libraries of veteran teachers and school administrators seeking a one stop coaching resource **Houghton Mifflin**

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from applications to skills **Mathematics** ,1991 **Prentice Hall Algebra 1** Jan Fair,1992

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the

two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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