

Covalent Bonding Practice

Instructions: Show how the following non metals will covalently bond using the steps below.

1. Draw the Lewis dot structure using the number of valence electrons for each element. *Note: You might want to use different colored pencils or dots and x's for different elements to help remember which electrons belong to which element.*
2. Rearrange the electrons in between the atoms so each can fulfil the octet rule. *Note: Hydrogen is an exception to the octet rule as it only needs 2 valence electrons to fulfil the octet to have the same electron configuration as helium.*
3. Circle each atom while making sure there are 8 electrons (or 2 for hydrogen) in each circle, which fulfills the octet rule. Circles will overlap.
4. Count the pairs of electrons that are shared in between each atom in the overlapping circles. There may be one pair, two pairs, three pairs or four pairs that are shared.
5. Redraw each atom and draw a line for each pair of electrons that are shared between the atoms.
6. Write the chemical formula using subscripts to indicate how many atoms there are of each element. *Note: There is no need to write a subscript if there is only one of an element (example: There are two hydrogens and one oxygen in H₂O.)*
7. Write the number of each type of covalent bond that is the atom makes depending on the number of electrons shared between each atom (1 pair=single, 2 pairs=double, 3 pairs=triple, 4 pairs=quadruple)

Example: Oxygen and Oxygen



2. Hydrogen and Hydrogen

Chemical Formula: _____

Type of Bond: _____

3. Chlorine and Chlorine

Chemical Formula: _____

Type of Bond: _____

4. Nitrogen and Nitrogen

Chemical Formula: _____

Type of Bond: _____

Lesson 8 2 Covalent Bonding Key

Leo J. Malone, Theodore O. Dolter



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Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 *Organic Chemistry Study Guide with Solutions Manual* Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer Primary Science: Knowledge and Understanding Graham Peacock, John Sharp, Rob Johnsey, Debbie Wright, 2024-04-14 All the subject knowledge you need to teach primary science The essential subject knowledge text for primary science Secure subject knowledge and understanding is the foundation of confident creative and effective teaching This comprehensive text covering the whole primary curriculum includes interactive tasks self assessment questions and links to other resources in all chapters Primary science matters This 10th edition includes links to the ITT Core Content Framework and new content on children s common misconceptions in science *Chemistry for the Utterly Confused* John T. Moore, Richard H. Langley, 2007-06-01 Banish bafflement in this tough subject From formulas and lab techniques to the periodic table Chemistry for the Utterly Confused focuses on the areas of maximum confusion and breaks down the most difficult chemistry topics into easy to understand concepts This invaluable guide also teaches problem solving skills you need to master this imposing subject Whether you re in high school in college or simply brushing up on chemistry knowledge this fun easily accessible book will make understanding chemistry a breeze Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text *Study Guide for Mortimer's Chemistry, Sixth Edition* Donald W. Shive, Louise E. Shive, 1986 **Introductory Chemistry** Kevin Revell, 2020-11-17 Introductory Chemistry creates light bulb moments for students and provides unrivaled support for instructors Highly visual interactive multimedia tools are an extension of Kevin Revell s distinct author voice and help students develop critical problem solving skills and master foundational chemistry concepts necessary for success in chemistry **Study Guide** Joan I. Senyk, Darrell D. Ebbing, 1984 *Bowker's Complete Video Directory 2001* ,2001 *Study Guide to Accompany Human Physiology* Kevin T. Patton, Victoria Schauf, 1989-12 *Books in Print Supplement* ,2002 *Introductory Chemistry* Charles H. Corwin, 2011 To help students learn chemical skills and concepts more

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