

## 7-1 Review and Reinforcement

### Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative	noble gas	octet	electron configuration
positive	electrons	ions	polyatomic
crisscross	anion	cation	empirical formula

- An ionic bond is an attraction between oppositely charged ions.
- Anions have a negative charge.
- An atom becomes an ion by losing or gaining electrons.
- The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
- When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
- The atoms that make up polyatomic ions are bonded together by covalent bonds.
- The crisscross method can be used to write the formula for an ionic compound.
- The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

- |                        |  |
|------------------------|--|
| <u>losing</u>          | 9. Calcium becomes a monatomic cation by <u>gaining</u> two electrons.                     |
| <u>true</u>            | 10. A cation has a <u>positive</u> charge.   |
| <u>true</u>            | 11. A binary ionic compound contains only <u>one</u> kind of cation and one kind of anion. |
| <u>polyatomic</u>      | 12. <u>Monatomic</u> ions consist of more than one atom.                                   |
| <u>seven</u>           | 13. The Lewis dot diagram for chlorine, a group 7A element, has <u>six</u> electrons.      |
| <u>true (released)</u> | 14. A <u>great deal of energy is produced</u> when an ionic compound is formed.            |

# Grade 7 Chapter Test Answer Keys Prentice Hall

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**Grade 7 Chapter Test Answer Keys Prentice Hall:**

The book delves into Grade 7 Chapter Test Answer Keys Prentice Hall. Grade 7 Chapter Test Answer Keys Prentice Hall is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Grade 7 Chapter Test Answer Keys Prentice Hall, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to Grade 7 Chapter Test Answer Keys Prentice Hall
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    - Chapter 4: Grade 7 Chapter Test Answer Keys Prentice Hall in Specific Contexts
    - Chapter 5: Conclusion
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  6. In chapter 5, the author will draw a conclusion about Grade 7 Chapter Test Answer Keys Prentice Hall. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Grade 7 Chapter Test Answer Keys Prentice Hall.

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