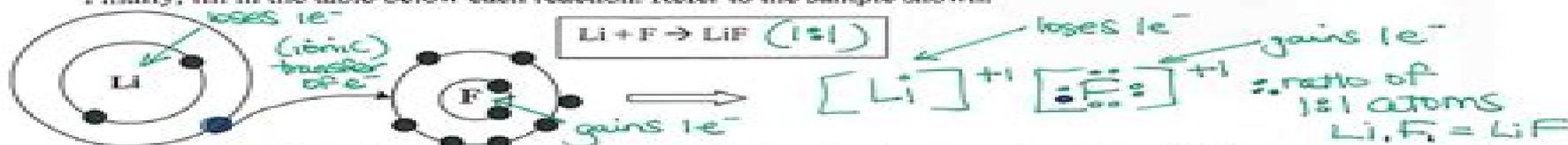


### Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1ve	transfers 1ve <sup>-</sup> to F	$\text{Li}^+$
F	7ve	accepts 1ve <sup>-</sup> from Li	$\text{F}^-$

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$ 	Li	1ve	donate $1e^-$	$\text{Li}^+$
	Cl	1ve	accepts $1e^-$	$\text{Cl}^-$
2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$ 	Ca	2ve	donates $2e^-$	$\text{Ca}^{+2}$
	O	6ve	accepts $2e^-$	$\text{O}^{2-}$
3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$ 	Be	2ve	donates $2e^-$ to each F	$\text{Be}^{+2}$
	F	7ve	each F accepts 1ve	$\text{F}^-$
4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$ 	Mg	2ve	donates $2e^-$ to S	$\text{Mg}^{+2}$
	S	6ve	accepts $2e^-$	$\text{S}^{2-}$
5) $\text{K} + \text{F} \Rightarrow \text{KF}$ 	K	1ve	donates 1ve <sup>-</sup>	$\text{K}^+$
	F	7ve	accepts 1ve <sup>-</sup>	$\text{F}^-$

# Ionic Bonds Section 7 2 Worksheet Answers

**Lei Huang**



**Ionic Bonds Section 7 2 Worksheet Answers:**

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