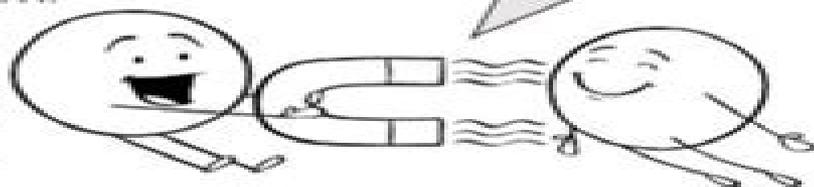


# CHEMICAL BONDS

## Ionic Bonding

Metals and non-metals react together to form \_\_\_\_\_ compounds. They are held together by ionic bonds. Ionic bonding is a force of \_\_\_\_\_ attraction between two \_\_\_\_\_ atoms (called \_\_\_\_\_). An atom becomes charged by either donating or accepting \_\_\_\_\_ electrons from another atom. A valence electron is one from the \_\_\_\_\_ shell of the atom.

Since opposites attract, there is an attraction between the positively charged and negatively charged ions that holds them together.



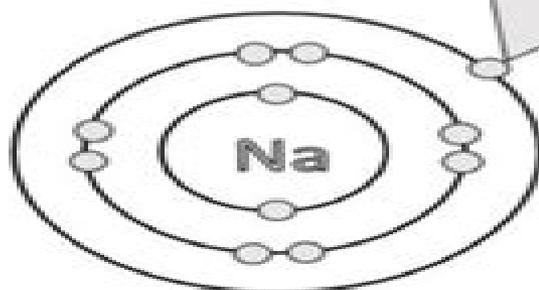
If an atom \_\_\_\_\_ an electron, it becomes positively charged. This is then called a \_\_\_\_\_ always donate electrons. If an atom accepts an electron, it becomes \_\_\_\_\_ charged. This is called an \_\_\_\_\_. Non-metals always accept electrons.

Atoms do this in order to become '\_\_\_\_\_'. A stable atom is one which has obtained a \_\_\_\_\_ of electrons.

*How do we know which atoms donate and which accept electrons?*

Sodium has \_\_\_\_\_ lone electron in the outer shell. It is easiest for this atom to \_\_\_\_\_ one electron than to try and find seven more to fill the outer shell.

Chlorine has \_\_\_\_\_ electrons in its outer shell. Therefore, it is easiest for Chlorine to \_\_\_\_\_ one electron than to try and donate seven.



All \_\_\_\_\_ donate electrons and therefore become cations.



All \_\_\_\_\_ accept electrons and therefore become anions.

# Ionic Bonds Worksheets For Middle School

**Arshad Iqbal**



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web september 11 2023 the new york public library for the performing arts from the sweet trippy melodies of the hippie era to the harsh yet experimental sounds of rock of the 1970s pink floyd s concept album the dark side of the moon released in march 1973 to an audience of young people disillusioned with contemporary society was a siren

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