

Chemistry: Ionic & Covalent Bonds Unit Study Guide

Covalent Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Covalent Compounds.

- Bondable
- Large / Small
- Atom / Molecule
- High / Low
- Conducts / Does not Conduct
- Solid / Liquid
- Soluble / Insoluble
- Variable / Constant

2. Write the prefix following number.

Number	Prefix
1	
2	
3	
4	
5	

3. What are the ions?

4. Complete the chart.

Formula	Name
Mg^{2+}	
	Br^{-}
Fe^{3+}	
	SO_4^{2-}

5. What does VSEPR

Name: _____ Date: _____ Period: _____

Ionic Bonding & Nomenclature Study Guide

1. Circle the correct word that would complete the statements below regarding Ionic Compounds.

- Bond between a Metal / Nonmetal and a Metal / Nonmetal
- Large / Small Electronegativity Difference
- The Metal / Nonmetal form Cations which Gain / Lose Electrons
- The Nonmetal / Nonmetal form Anions which Gain / Lose Electrons
- High / Low Melting & Boiling Point
- Conducts / Does not Conduct Electricity in Solution
- Solid / Liquid - Good at Room Temperature

2. What charge would the following periodic table families have in a chemical bond?

Family	Charge
Alkali Metals	
Alkali Earth	
Halides	
Carbon	
Nitrogen	
Oxygen	

3. Write the number numbers that is represented by the following numbers.

Number	Prefix
1	
2	
3	
4	
5	
6	

4. Complete the chart below by filling in the missing names.

CATION	ANION	FORMULA	NAME
Mg^{2+}	Br^{-}		
		CuO	
			Iron (III) sulfate
		$(NH_4)_3PO_4$	

Ionic Covalent Bonding Test Study Guide Answers

CH Cherryholmes



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