

MEDIC 2024 EXAM 1 review

Covalent bond- force holding atoms together that results from atoms sharing electrons

Protons- positively charged particles within nucleus of an atom

Electrons- negatively charged particles that orbit the nucleus of an atom

Molecule- smallest fundamental unit of a chemical reaction made up of atoms held together by one or more covalent bonds.

Isotopes- variants of the same elements having the same # of protons but varying in number of neutrons.

***Ion-** charged particle

***Anion** ion with a negative charge will be attracted to anode or positive pole

***Cations** ion w positive charge will attract to cathode or negative pole
most abundant intracellular is Potassium

Hydrogen bond- weak bond formed by the attraction between a slightly positively charged hydrogen atom and a slightly negatively charged oxygen atom (HO₂)

Ionic bond- chemical bond resulting from attraction between an atom or molecule with negative charge and an atom or molecule with a positive charge

***Cell-** basic structural unit of all plants and animals. Membrane enclosing a thick fluid and a nucleus. Membrane is selective barrier that keeps good in and bad out. Contains Fats, proteins and CHO

***Nucleus-** brains of the cell contains DNA & RNA. In cells of higher organisms nucleus is surrounded by membrane DNA tell cbody what to do

***Mitochondria-** powerhouse of the cell. Principle site of the conversion of food into energy. Makes ATP which is source of energy, 1 mole of sugar produce 32-28 ATP in aerobic metabolism. And only 6-7 in anaerobic (plus lactic and oyruvic acids)

***ATP-** Adenosine Triphosphate- ATP- high energy compound present in all cells esp muscle cells. When split with enzymes ATP yields energy. FUEL OF THE CELL

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S Marginson



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