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labflow



Figure 4 Boiling water bath setup

Objectives

You will prepare a calibration curve for your thermometer using liquid. Before preparation of the data, the value of the boiling point must be adjusted for atmospheric pressure. Although both melting points and boiling points change with varying pressure, the dependence of the melting point on pressure is very small and negligible. However, the rate of change of the boiling point with pressure is large. Atmospheric pressure at sea level is about 1013 mbar. The boiling point of water at 1013 mbar is equal to 100°C. During the graph, you will be told the atmospheric pressure will be slightly lower than 1013 mbar. The atmospheric pressure on your report sheet during process. Please remember the boiling point of water at sea level is 100°C. The pressure on the boiling point can be estimated by assuming 1013 mbar. The result can vary due to atmospheric pressure is below 1013 mbar. The

P. atmospheric = (1013 mbar) - (atmospheric pressure) = 0.001 $\frac{mbar}{mbar}$ (1.0)

is subtracted from 1013 mbar to get the boiling point of water at the process.

P. atmospheric = (1013 mbar) - (1013 mbar) = 0.001 $\frac{mbar}{mbar}$ = 0.001

The boiling point at 1013 mbar would be 100.00°C = 1.00°C = 100.00°C.

28. Change the axes titles in the spreadsheet to the appropriate units. The x-axis is labeled "Boiling Point" and the y-axis is labeled "Pressure (mbar)".

Click on the title of the graph and change it to your name. Please add the name of your graph at the end of your name.



29. Print out the graph and attach it to the papers that need to be handed over work.

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