

Name: _____

The Hydrogen Atom – Student Guide

Background Material

Carefully read the background pages entitled Energy Levels, Light, and Transitions and answer the following questions to check your understanding.

Question 1: Complete the following table which compares how the Bohr Model and the Quantum model represent the Hydrogen atom. In some cases they both describe things in the same way and in some cases they do not.

Concept	Bohr Model	Quantum Model
Nucleus	positively charged proton at the center	positively charged proton at the center
Electrons		negatively charged cloud
Orbital		
Location of electron in the first orbital	always located at the Bohr radius	
Energy of the Energy Levels		$E = -E_0/n^2$

Question 2: Complete the following table which relates the parameters of two different photons by circling the appropriate relationship. The first row is completed for you: "A red photon has a larger wavelength, smaller frequency, smaller energy, and the same velocity through space as a blue photon".

Hydrogen Atom Student Guide Naap Answers

**Rudolph M. Langer, California Institute
of Technology. Division of Physics,
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Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text *Student's Guide to Chemistry; a Modern Introduction* David Brooks, 1974 [Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e](#) Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry *On the Energy Levels of a Model of the Compressed Hydrogen Atom* S. R. de Groot, C. A. ten Seldam, 1946 **The Hydrogen Atom Energy Levels**, 1976 **Student Solutions Manual for General Chemistry** John McMurry, Robert Fay, 2013-03-07 By Joseph Topich Virginia Commonwealth University This manual for students contains solutions to selected all in chapter problems and even numbered end of chapter problems *Student Study Guide/solutions Manual to Accompany Foundations of Inorganic, Organic, and Biological Chemistry* Katherine J. Denniston, 1995 This student study guide has been written to accompany Foundations of Inorganic Organic and Biological Chemistry by Caret Denniston and Topping It was designed to complement the text not to be used in place of the text Each chapter of the study guide contains the following sections reorganized and expanded set of learning goals concise chapter summary in chapter solved problems list of key terms review problems Preface [The Hydrogen Atom](#) Rudolph M. Langer, California Institute of Technology. Division of Physics, Mathematics and Astronomy, 1927 **The Electron and Proton Orbits in the Hydrogen Atom** H. Vic Dannon, 2021

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