

[4317] - 5
F.Y. B.Sc.
CHEMISTRY - I
Physical and Inorganic Chemistry
(Paper - I) (2008 Pattern) (Theory)

Time : 3 Hours]

[Max. Marks : 80

Instructions to the candidates :

- 1) All questions are compulsory.
- 2) Draw neat diagrams wherever necessary.
- 3) Figures to the right indicate full marks.

Q1) Answer the following questions :

[16]

- a) Find the value of y in the equation $y = 10^{-6.2218}$.
- b) Define the term surface tension.
- c) What is the effect of temperature on viscosity?
- d) Explain the term catalyst with example.
- e) State second law of thermodynamics in the form of entropy.
- f) BeF_2 is linear but SF_2 is angular. Why?
- g) How many moles are present in 90 grams of water.
- h) What are the types of hydrogen bond.

Q2) a) Attempt any four of the following :

[8]

- i) Calculate the pH of $1.2 \times 10^{-3} \text{M HCl}$.
- ii) Find the value of x if $\log_e 6.321 = x$.
- iii) State any two rules of derivative.
- iv) Find the equation of line passing through the two points $(-1, -1)$ and $(2, 6)$.
- v) Arrange the following equation in the form of $y = mX + C$ and find out slope and intercept.
 $15y = X + 30$.
- vi) If $y = \frac{x}{x^2 + 1}$, find $\frac{dy}{dx}$.
- vii) If $y = (3x^2 + 1)(x^3 + 2x)$; find $\frac{dy}{dx}$.
- viii) $\int_2^8 x^3 \cdot dx = ?$

Fybsc Chemistry Question Paper

A. L. H. SMITH



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MCQs Chemistry V. K. Ahluwalia, 2013-12-30 This book has been divided in 37 chapters under three parts Inorganic Chemistry Organic Chemistry and Physical Chemistry for convenient understanding It also includes solved model test papers of the previous three years of AIIMS CBSE PMT CPMT UP to enable students to develop the skills of problem solving and time management essential for any entrance examination In addition to providing answers to all the questions detailed explanatory notes to selected difficult questions have also been provided to justify the answer A separate section of Assertions and Reasons is also given at the end of each chapter Exhaustive Question Bank Explanatory Notes and Hints Assertions Reasons Includes Pre solved papers of five years Models Test Papers of AIIMS CBSE PMT CPMT B.Sc. Sem.1 & 2 and 1 year Chemistry old question papers PRAHSIDDH Educare, 2024-05-28 Examination conducted by Bhavnagar University like B Sc Chemistry There is a set of old question papers of semester 1 and 2 as well as first year **New 21st Century Chemistry** L. H. M. Chung, 2016 **College Chemistry Questions and Answers PDF** Arshad Iqbal, The College Chemistry Quiz Questions and Answers PDF Class 11 12 Chemistry Competitive Exam Questions Chapter 1 6 Practice Tests Grade 11 12 Chemistry Textbook Questions for Beginners includes revision guide for problem solving with hundreds of solved questions Class 11 12 Chemistry Questions and Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry Quiz PDF book helps to practice test questions from exam prep notes The Grade 11 12 Chemistry Quiz Questions and Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved tests Class 11 12 Chemistry Questions and Answers PDF Free download chapter 1 a book covers solved common questions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Chemistry Interview Questions and Answers PDF Download free eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The Class 11 12 Chemistry Interview Questions Chapter 1 6 PDF book includes college question papers to review practice tests for exams Class 11 12 Chemistry Practice Tests a textbook s revision guide with chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Questions Bank Chapter 1 6 PDF book covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure Questions Chapter 2 Basic Chemistry Questions Chapter 3 Chemical Bonding Questions Chapter 4 Experimental Techniques Questions Chapter 5 Gases Questions Chapter 6 Liquids and Solids Questions The Atomic Structure Quiz Questions PDF e Book Chapter 1 interview questions and answers on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model defects charge to mass ratio of electron discovery of electron discovery of neutron discovery of proton dual nature of matter electron charge electron distribution electron radius and energy derivation electron velocity electronic configuration of elements energy of revolving electron fundamental

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block between the front engine support and the starter 3ya1 3yc1

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